

FIG. 1

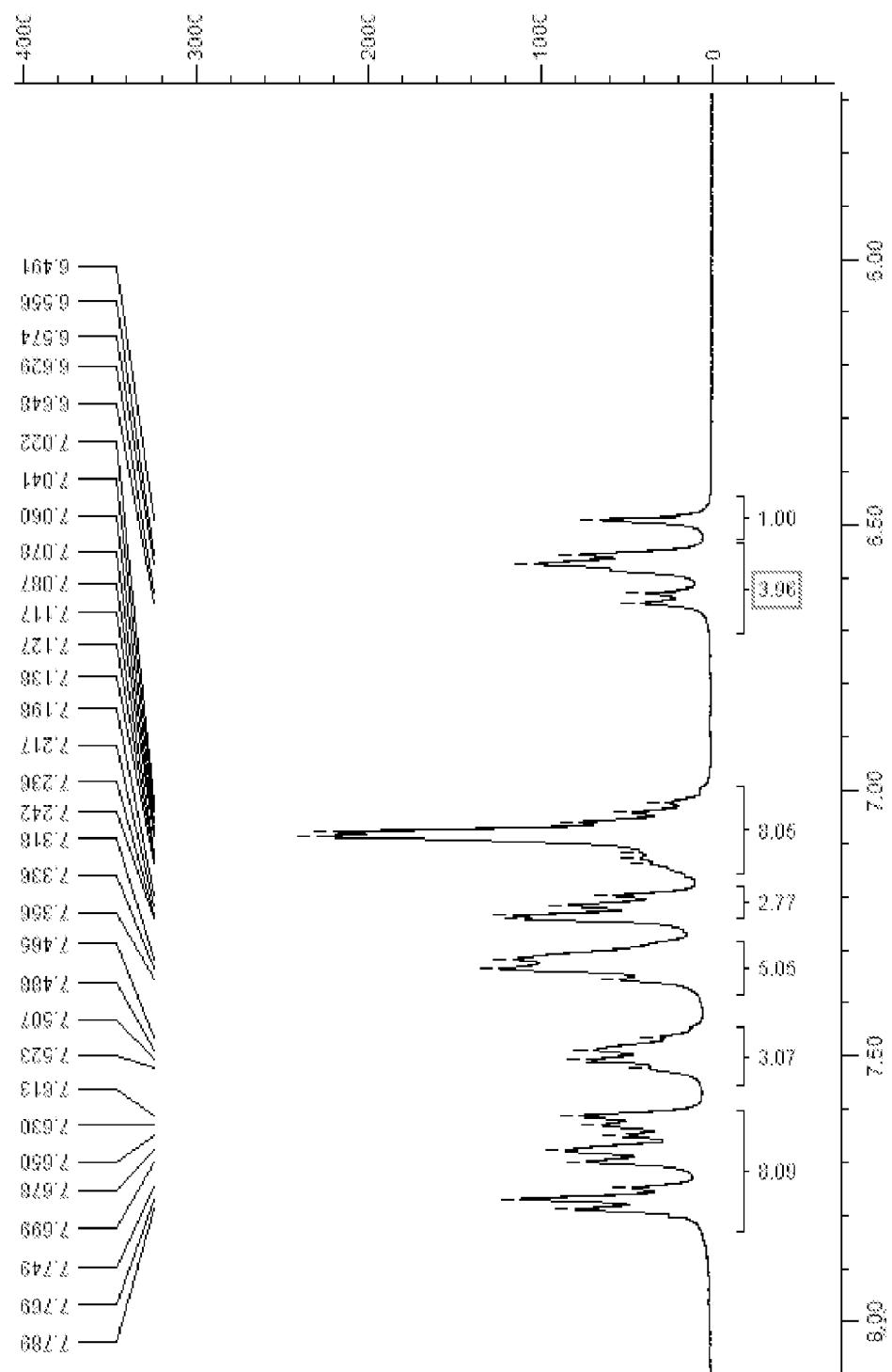


FIG. 2

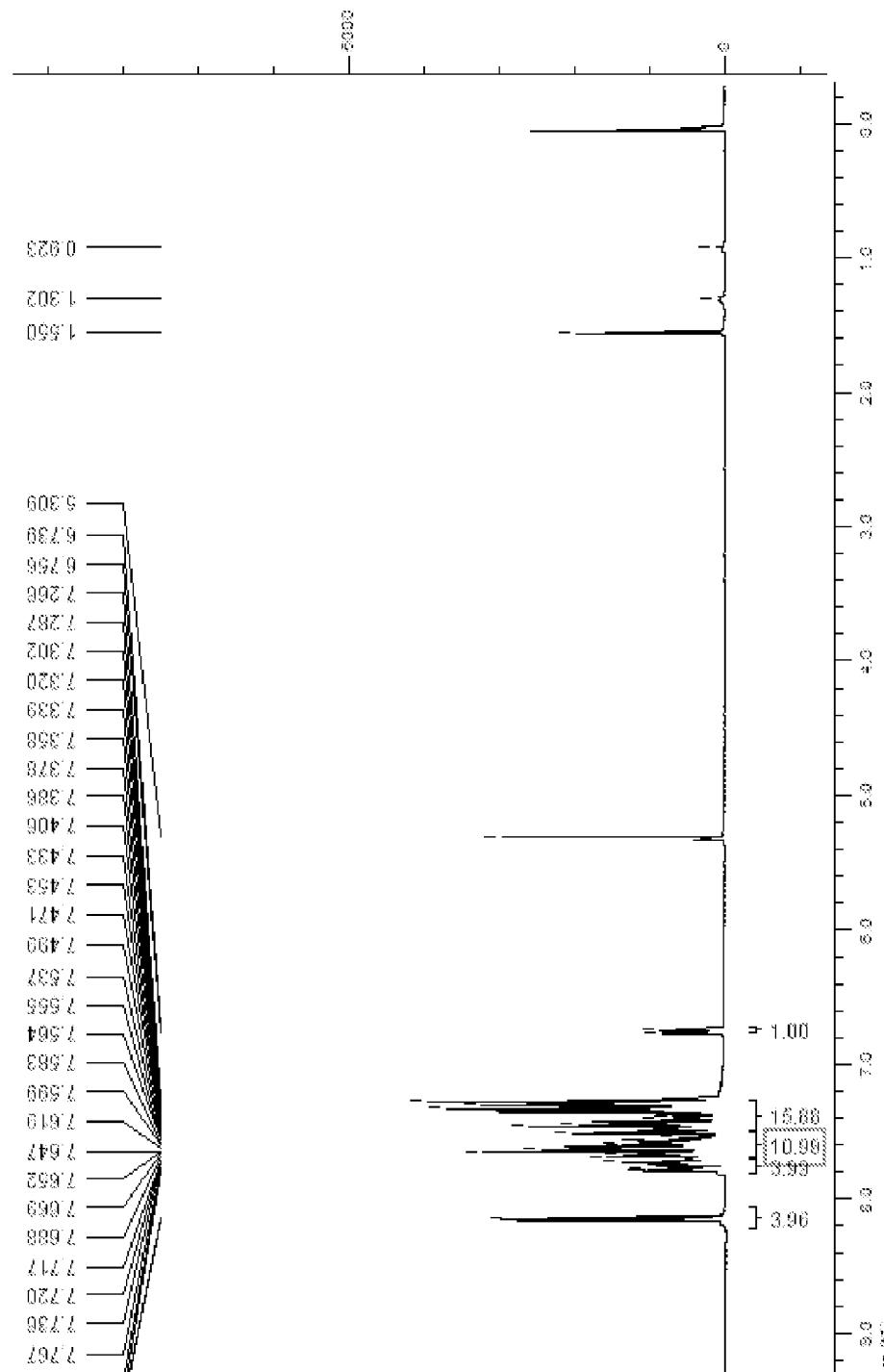


FIG. 3

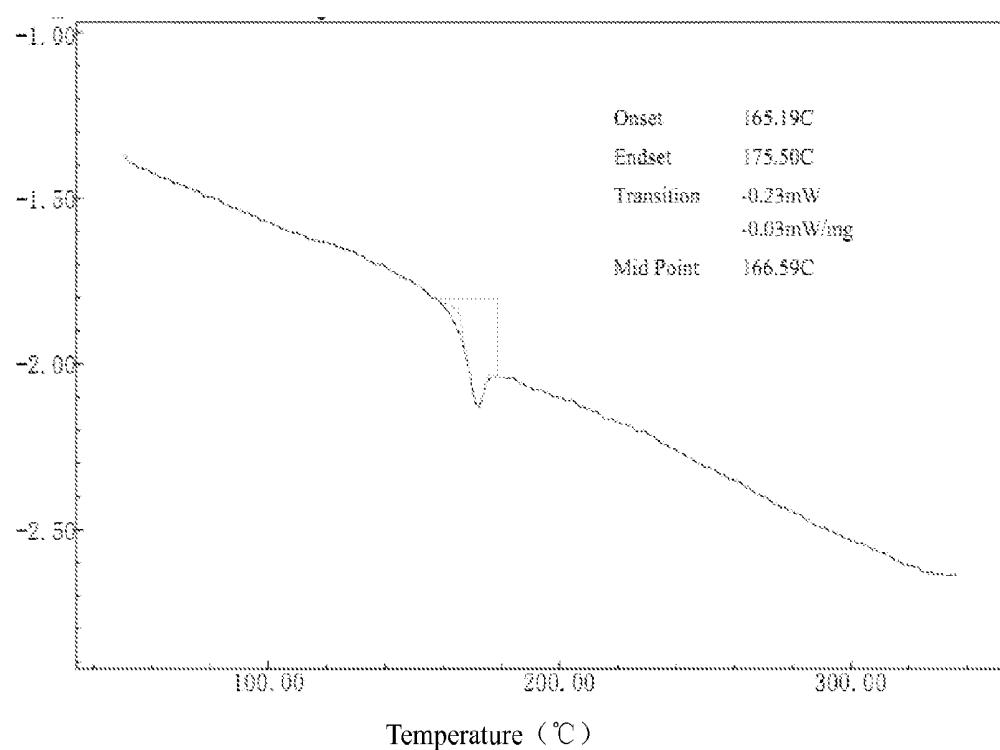


FIG. 4

## ORGANIC ELECTRONIC MATERIAL AND ORGANIC ELECTROLUMINESCENT DEVICE

### TECHNICAL FIELD

**[0001]** This invention relates to a new type of organic electronic material. By vacuum deposition into thin film, it is used in organic light emitting diodes as an electron transport material and phosphorescence host material. This invention belongs to the organic light-emitting device (OLED) display technical field.

### BACKGROUND ART

**[0002]** OLED, as a new type of display technology, has unique advantages such as self-illumination, wide viewing angle, low power consumption, high efficiency, thin, rich colors, fast response, extensive application temperature range, low driving voltage, applicable for flexible and transparent display panel, and environmental friendliness, etc. Therefore, OLED technology can be applied to flat panel displays and new generation of lighting, or can be used as backlight of LCD. In 1987, Kodak (Tang et al) firstly made a sandwiched bilayer devices using 8-hydroxyquinoline aluminum ( $\text{Alq}_3$ ) as the light emitting layer and a triphenylamine derivative as the hole transporting layer through thin-film vacuum evaporation technique. The device achieved a luminance up to  $1000 \text{ cd/m}^2$  at a driving voltage of 10 V (Tang C. W., Vanslyke S. A. *Appl. Phys. Lett.* 1987, 51, 913-916). This technological breakthrough had aroused widespread notice in the scientific community and industry, and since then the research and applications of organic light-emitting technology become a hot issue. Subsequently, in 1989, with the invention of host-guest emission technology, the luminous efficiency and lifetime of OLED had been greatly improved. In 1998, Forrest et al found the electrophosphorescence, which lifted the theoretical limit of organic electroluminescent quantum efficiency from 25% to 100% (Baldo M. A., Forrest S. R. *Et al, Nature*, 1998, 395, 151-154), bringing the research of organic light-emitting technology into a new era with extended field of research.

**[0003]** A classic three-layer OLED comprises a hole transport layer, a light emitting layer and an electron transport layer. adopts traditional electron transport material is  $\text{Alq}_3$ , which has good film-forming property and thermal stability, but its strong green emission and low electron mobility restricts its industrial applications. Subsequently, some electron transport materials with excellent performance, such as TPBI, BCP, Bphen, are also widely used in the OLED. The existing material of light-emitting layer can be divided into two categories, namely, fluorescent material and phosphorescent material, which are often adopted in guest-host doping technology.

**[0004]** CBP (4,4'-bis(9-carbazolyl)-biphenyl) is a highly efficient and high-triplet-energy phosphorescent host material. When CBP is used as the host material, triplet energy can be smoothly transferred to phosphorescent material, producing efficient red and green emission. However, such representative host materials are restricted to use because of their poor thermal stability and short lifetime of manufactured devices.

**[0005]** Although OLED has made considerable progress after 20 years of development, and organic materials have also been in constant development, there are still very few

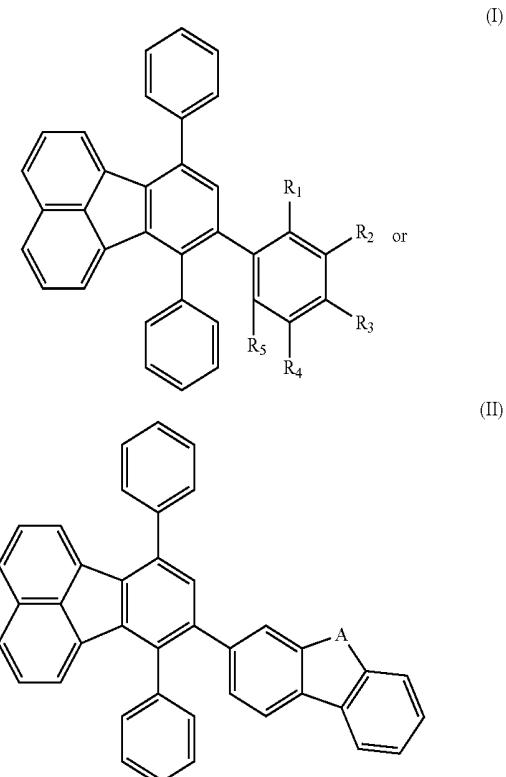
materials that can meet the market demands with good device efficiency and lifetime to give excellent performance and stability at the same time.

**[0006]** Fluoranthene, as an electroluminescent material, is widely used as electron transport material, hole transport material and light-emitting material. However, according to the known reports, either the performances of devices are not described in details, or the devices only had low efficiency or poor stability. In the present invention, a series of new compounds based on fluoranthene are disclosed and used in OLEDs.

### SUMMARY OF THE INVENTION

**[0007]** The purpose of the present invention is to provide the synthesis method of a novel organic compound which is used in electroluminescent devices as an electron-transport and/or a phosphorescent host material, and the preparation method of high performance OLEDs using the invented material hereof.

**[0008]** The organic electronic material in the present invention has a chemical structural formula (I) or (II).



**[0009]** wherein, any four of  $R_1$ - $R_5$  are hydrogen, and the other is C1-C6 alkyl, C7-C24 aryl, substituted or unsubstituted heteroaromatic ring(s) having one or more heteroatoms (N, O, S), triaryl silyl, diarylamino, diaryl phosphine oxide, aromatic carbonyl, arylthio; or  $R_1$ ,  $R_1$ ,  $R_3$ ,  $R_5$  are hydrogen,  $R_2$ ,  $R_4$  independently represent C1-C6 alkyl, C6-C24 aryl, heteroaromatic ring(s) with one or more heteroatoms (N, O, S), triaryl silyl, diarylamino, diaryl phosphine oxide, aromatic carbonyl, arylthio,

[0010] A represents  $C(R_6)_2$ ,  $N(R_7)$ , S, O,  $P(R_8)_2$ ,  $S(O)_2$  or  $B(R_9)$ ,  $R_6$ - $R_9$  independently represent hydrogen, deuterium, alkyl, phenyl, alkylphenyl, heteroaromatic ring with one or more heteroatoms (N, O, S), a cyclized structure formed between two  $R_6$  and C.

[0011] Preferably,  $R_1$ - $R_5$  are five-membered or six-membered heteroaromatic ring substituted with fluorenyl or alkyl or phenyl or naphthyl, diphenyl amino, phenyl-naphthylamine, triphenylsilyl, diphenyl phosphine oxide, phenyl carbonyl or phenyl sulfonyl.

[0012] Wherein, the said five-membered or six-membered heteroaromatic ring is carbazolyl, pyrimidinyl, pyridyl, thiazolyl, triazole group or triazine group.

[0013] The said fluorenyl is 9,9-dimethyl-fluorenyl, 9,9-diphenyl fluorenyl, 9,9-di-p-tolyl-9H-fluorenyl or spiro fluorenyl.

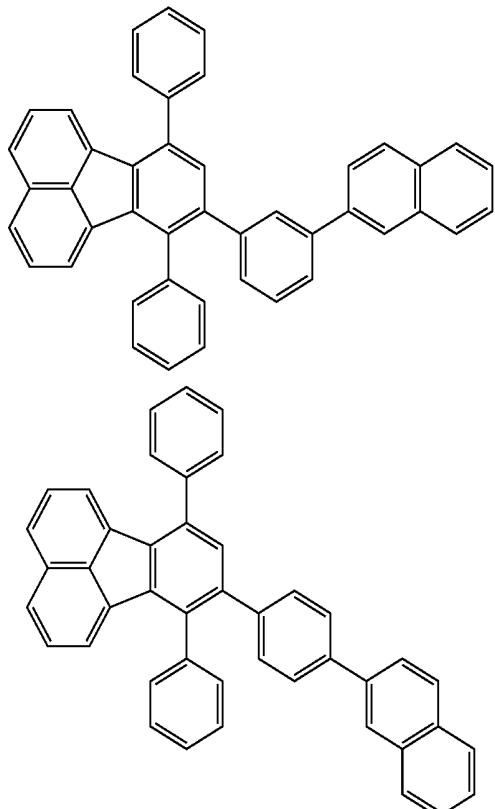
[0014] In one aspect, the said  $R_1$ ,  $R_2$ ,  $R_4$ ,  $R_5$  are hydrogens.

[0015] Preferably, A is  $C(R_6)_2$ ,  $N(R_7)$ , S, O,  $S(O)_2$ ;  $R_6$ - $R_7$  are hydrogen, methyl, phenyl, methyl phenyl, or five-membered cyclic structure formed between two  $R_6$  and C.

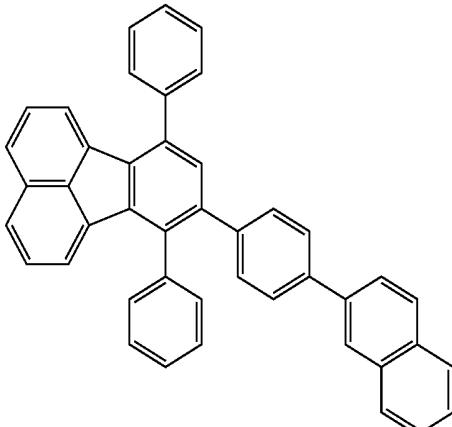
[0016] Preferably the compounds in the present invention are exemplified as but not limited to the follows:

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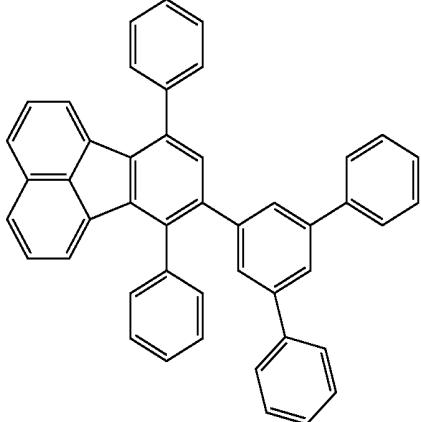
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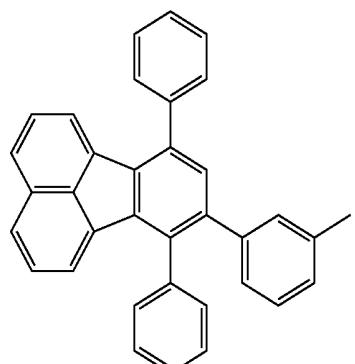
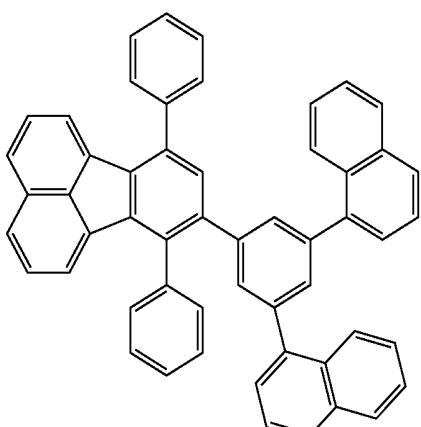
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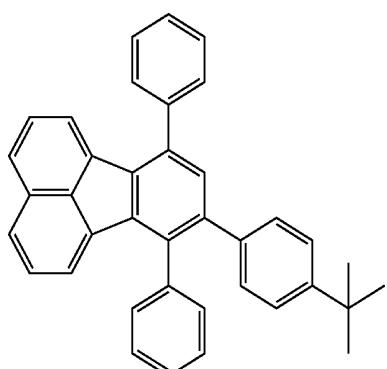
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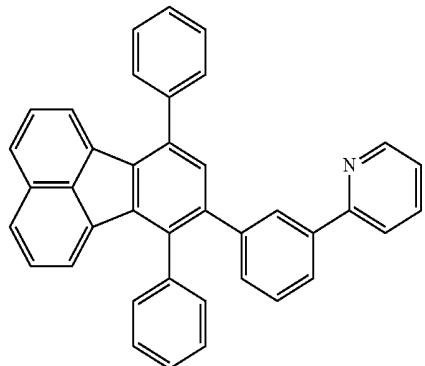
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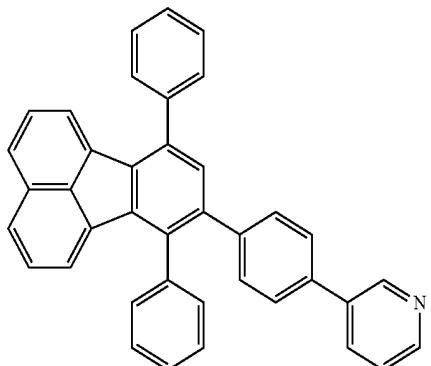
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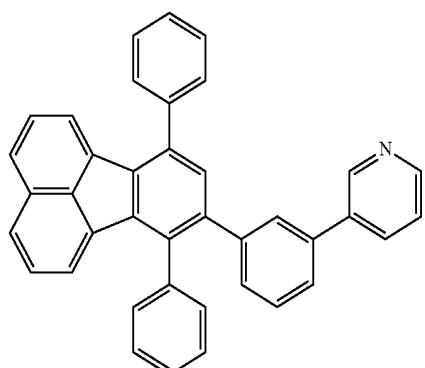


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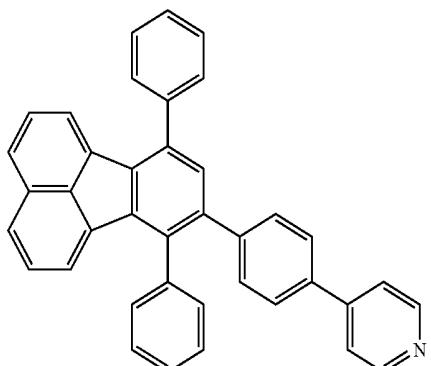
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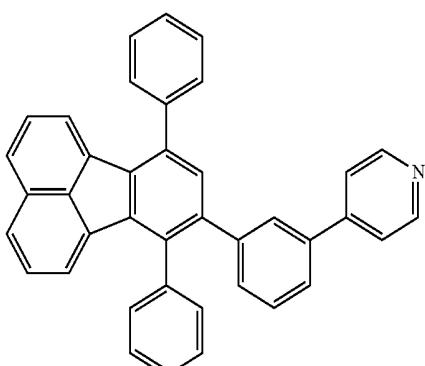
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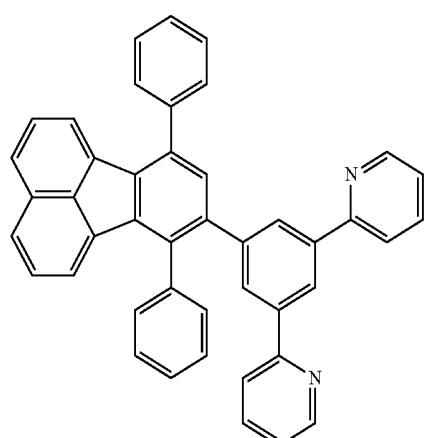
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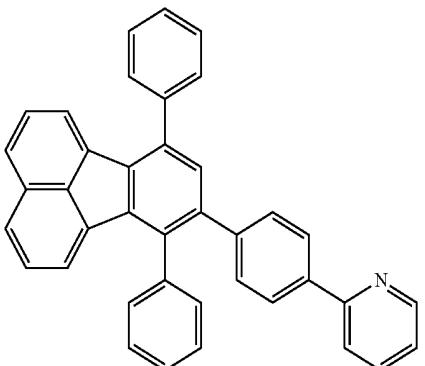
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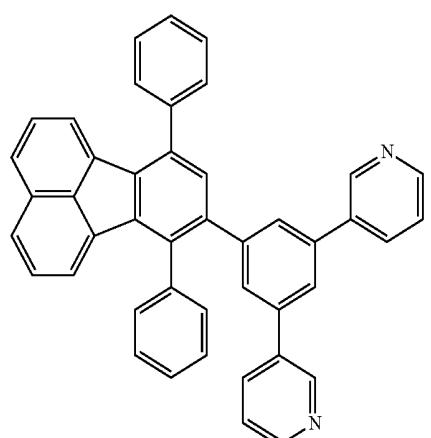
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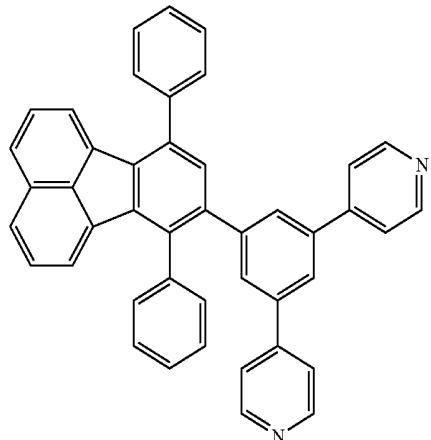


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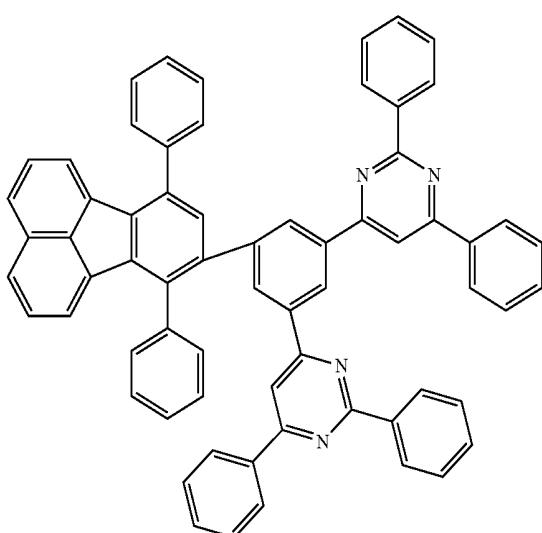
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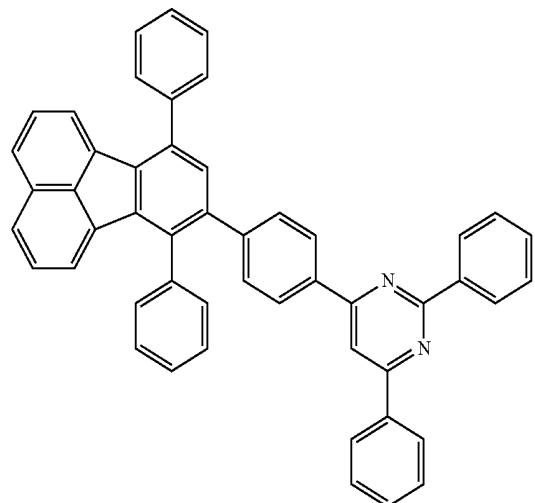


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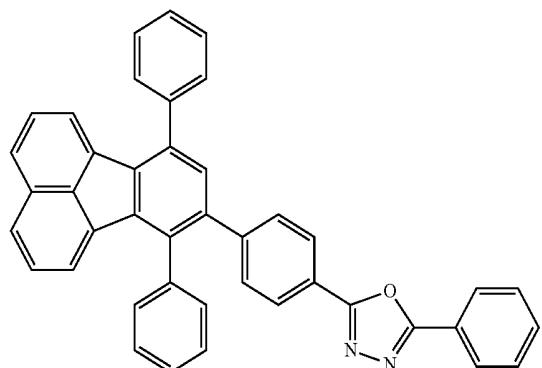
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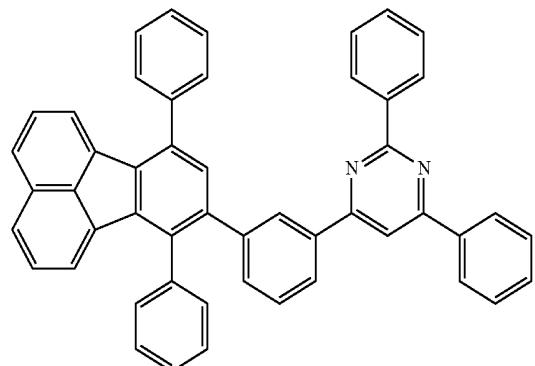
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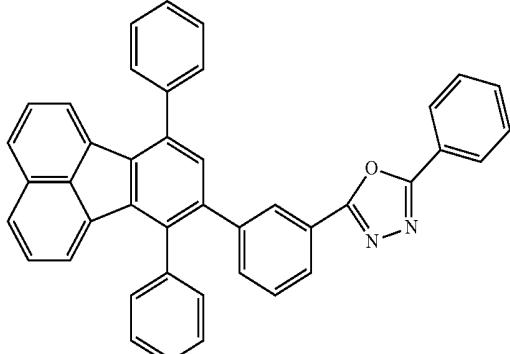
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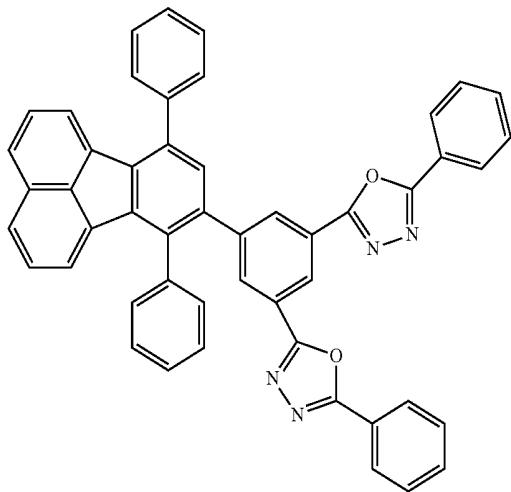


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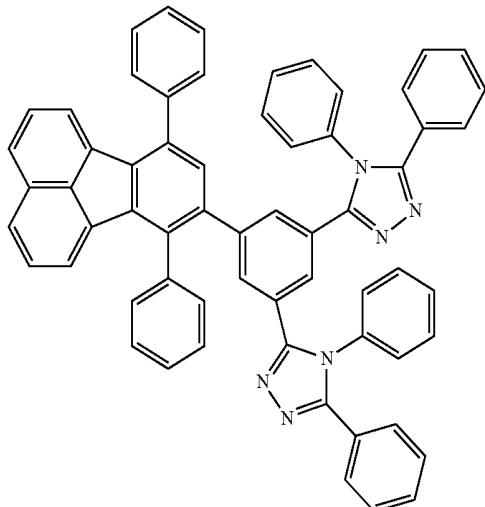
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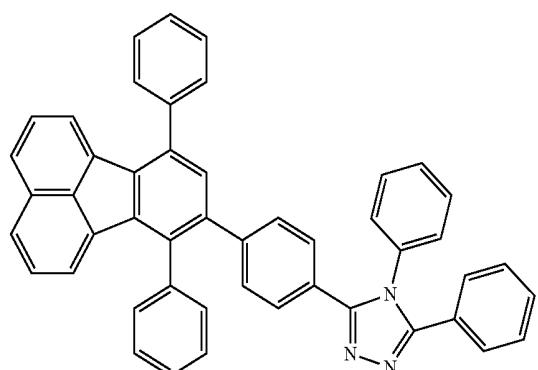


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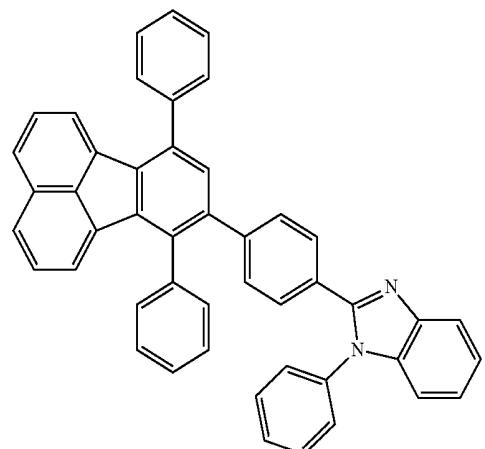
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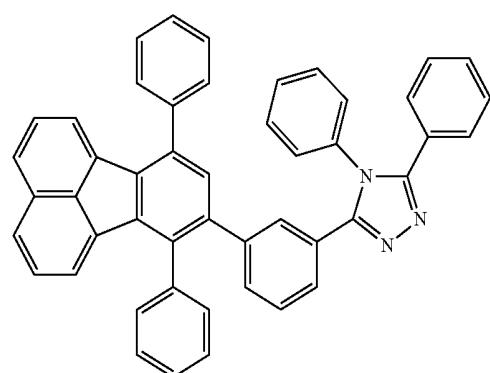
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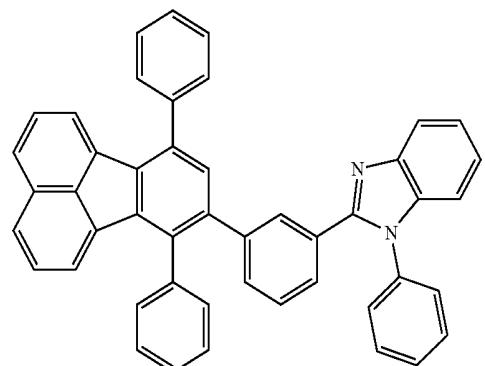
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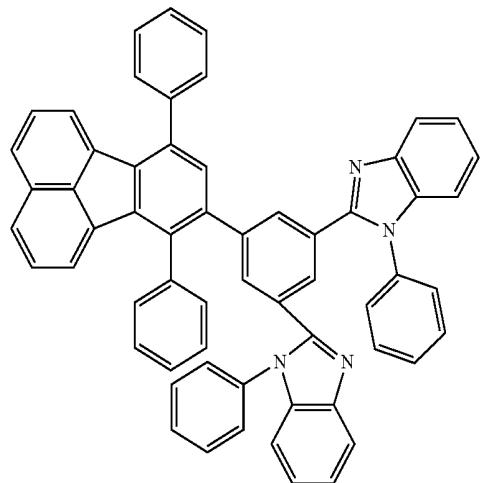
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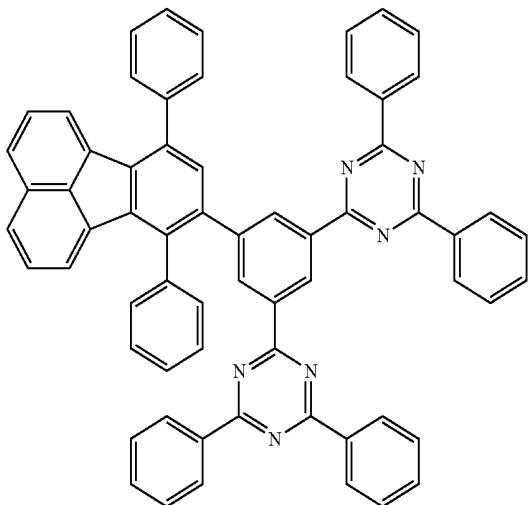


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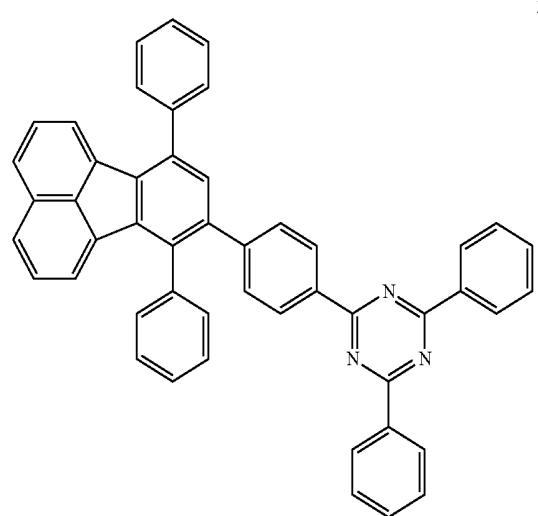
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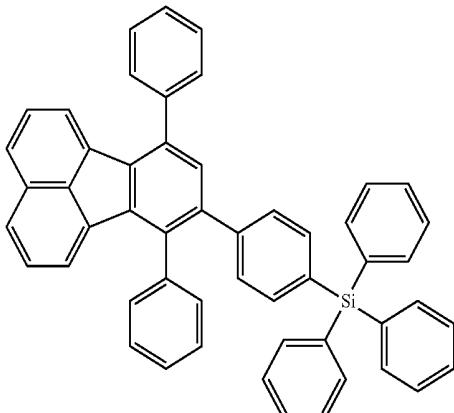


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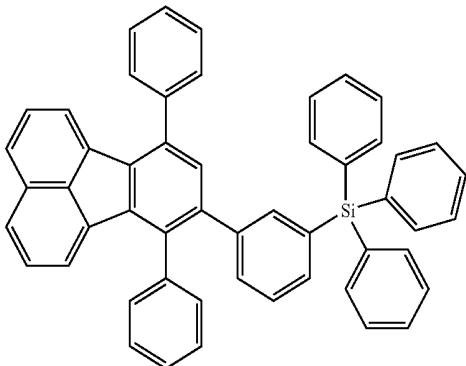
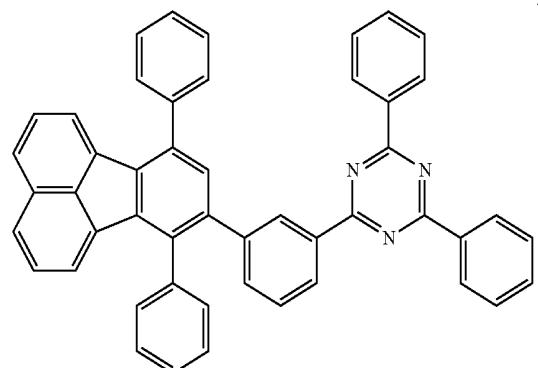
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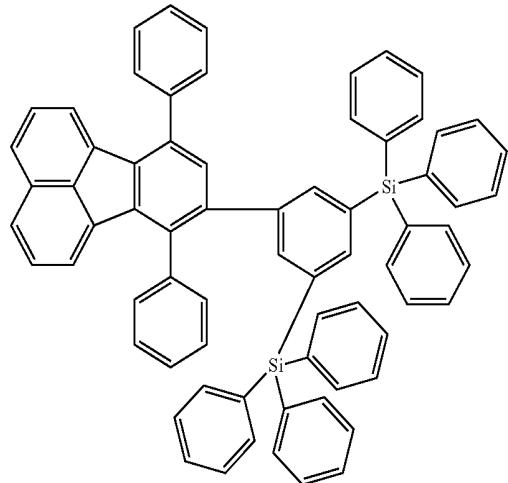


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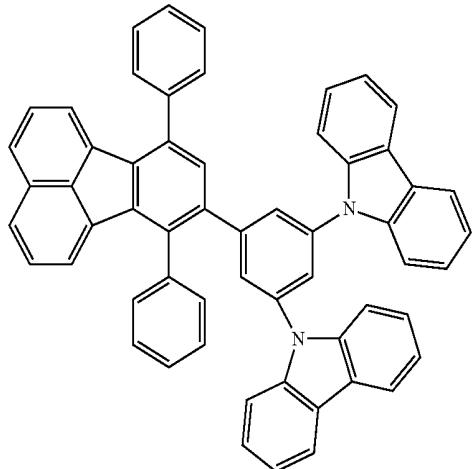
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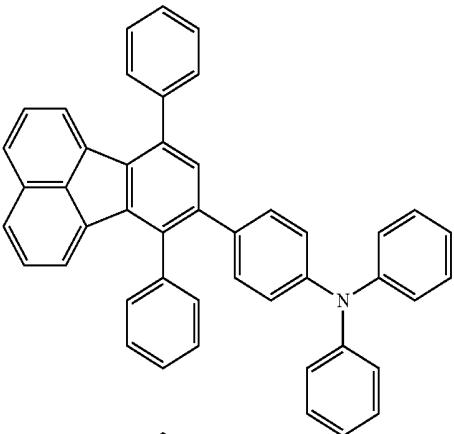


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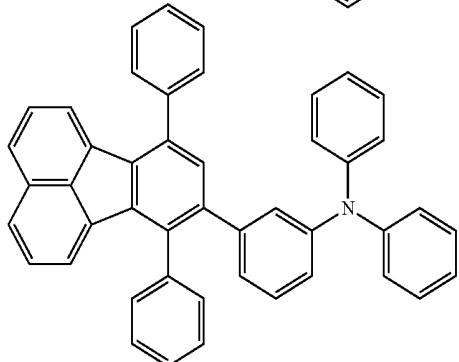
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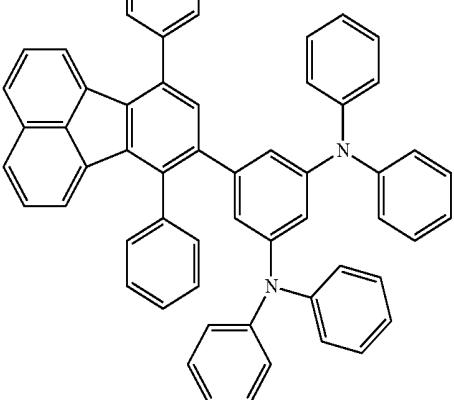
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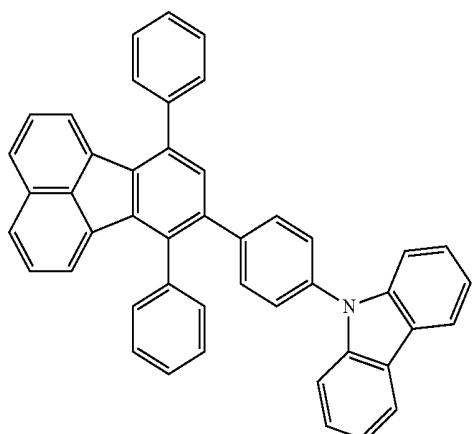
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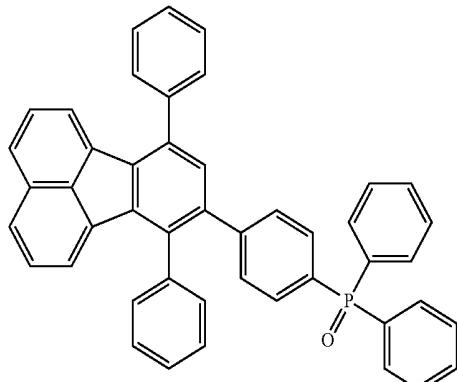
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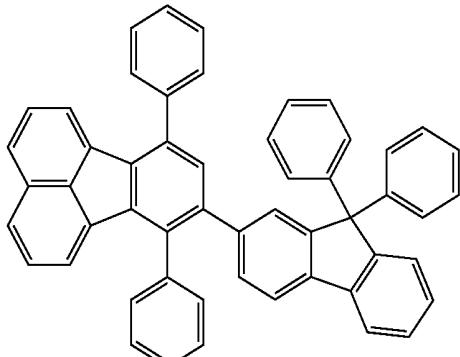


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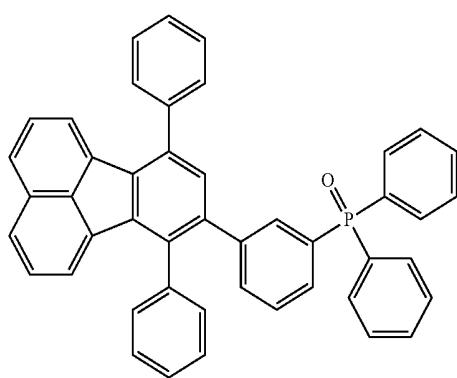
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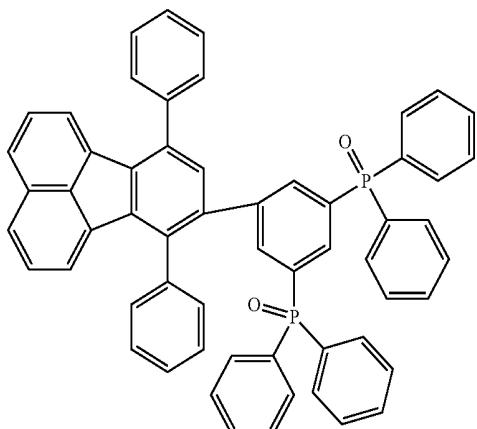
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The chemical structure is a bis(anthracene) molecule. It features two fused anthracene units. A central biphenyl group is attached to the 2 and 7 positions of each anthracene ring. Two triphenylmethyl (trityl) groups are attached to the 9 and 9' positions of the biphenyl group, with each trityl group consisting of a central phenyl ring substituted with three phenyl groups.

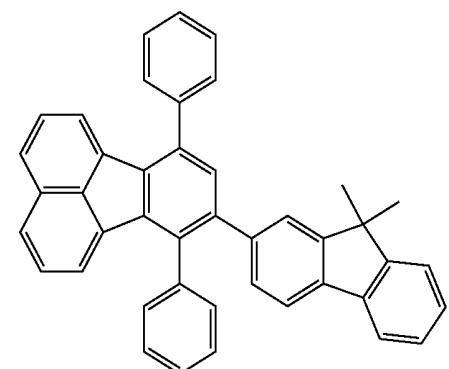
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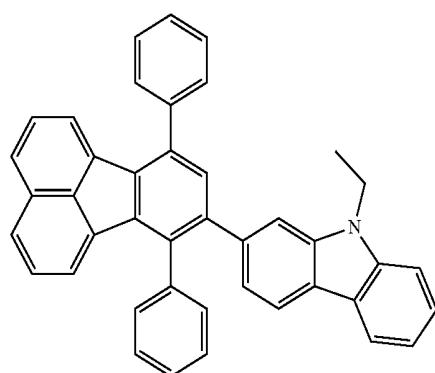


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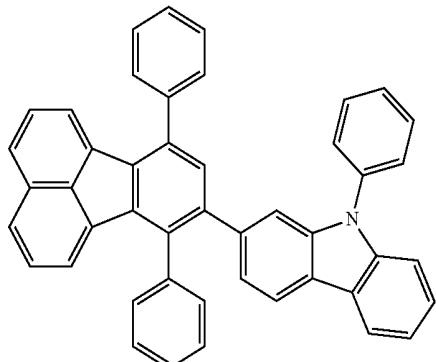
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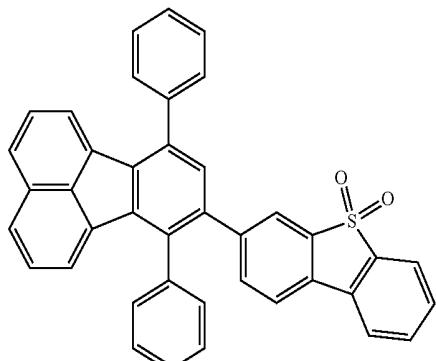
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[0017] More preferably:

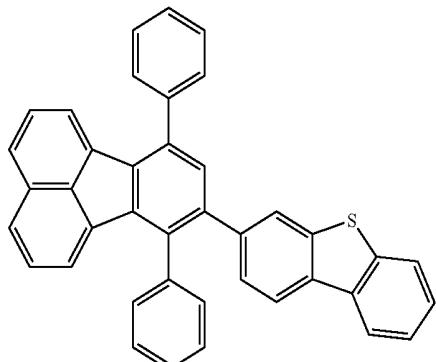
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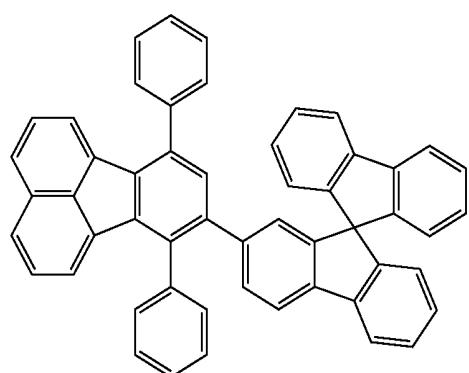
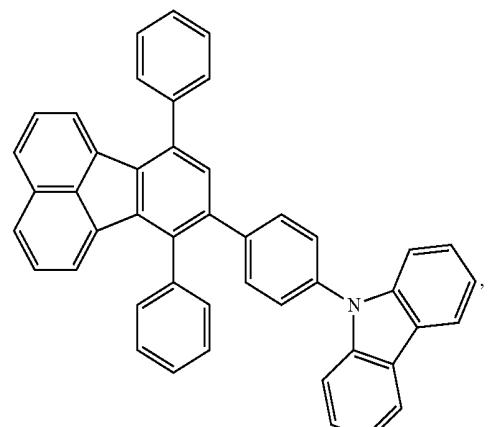
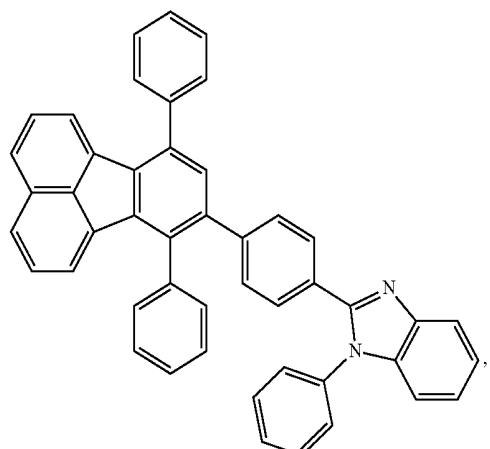
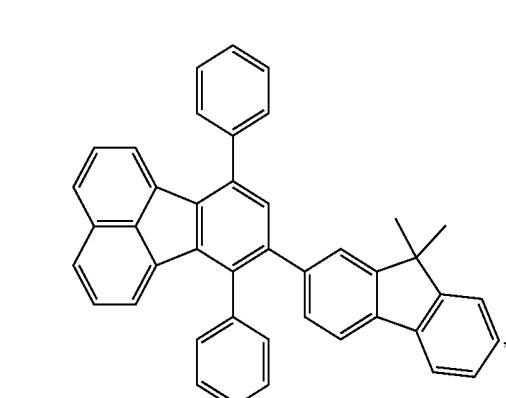
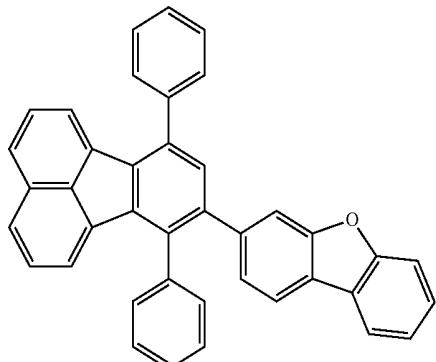
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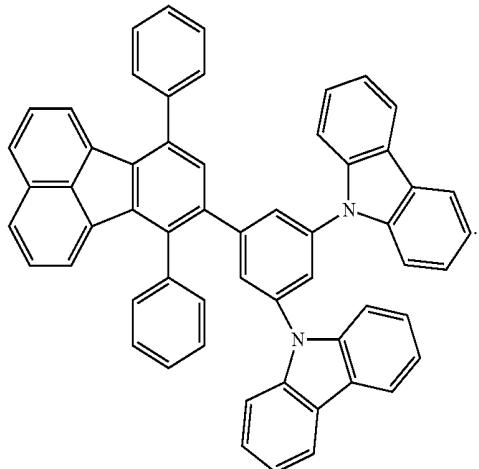
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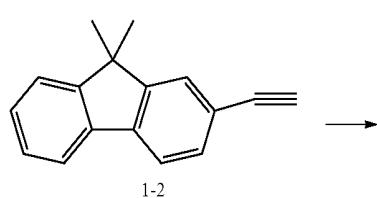
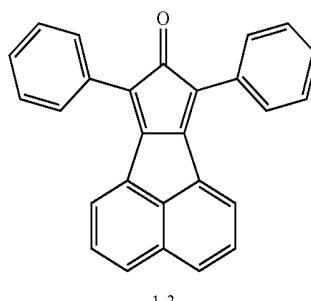
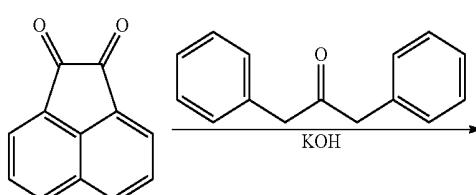
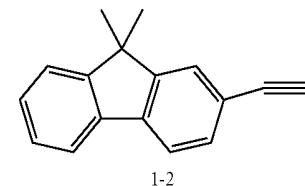
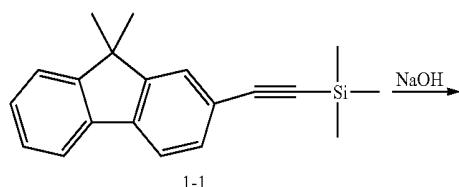
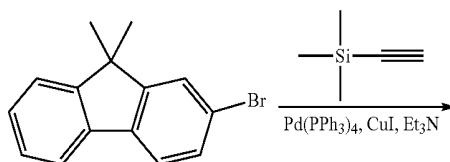
## DETAILED DESCRIPTIONS OF EMBODIMENTS

[0028] In the following, the present invention is described in details by the following embodiments.

## Embodiment 1

## Synthesis of Compound 43

[0029]



## BRIEF DESCRIPTION OF THE DRAWINGS

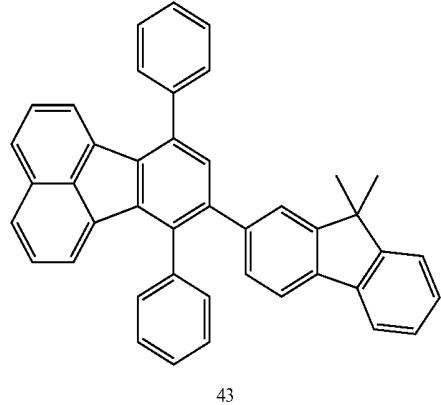
[0024] FIG. 1 is a structural drawing of the device, of which, 10 denotes a glass substrate, 20 denotes an anode, 30 denotes a hole injection layer, 40 denotes a hole transport layer, 50 denotes a light emitting layer, 60 denotes an electron transport layer, 70 denotes an electron injection layer, 80 denotes a cathode.

[0025] FIG. 2 is the  $^1\text{H}$  NMR diagram of compound 46.

[0026] FIG. 3 is the  $^1\text{H}$  NMR diagram of compound 36.

[0027] FIG. 4 is the DSC diagram of compound 46.

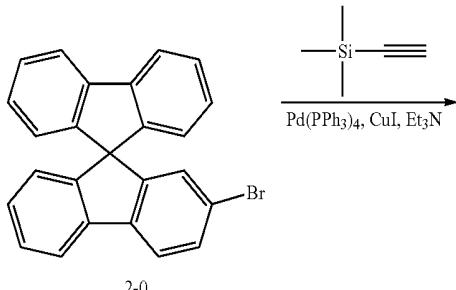
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Embodiment 2

Synthesis of Compound 46

[0034]



Synthesis of Intermediate 1-1

[0030] Triethylamine was added to a mixture of 7.5 g 2-bromo-9,9-dimethyl-fluorine, 0.15 g cuprous iodide, and 0.3 g tetrakis(triphenylphosphine)palladium, under nitrogen. The starting materials were stirred until fully dissolved, and then 10 ml trimethylsilyl acetylene was added followed by refluxing for 12 hours overnight. The solvent was removed under reduced pressure, the residue was extracted three times using diethyl ether/water. The organic phase was washed three times with saturated brine, dried over sodium sulfate, filtered, and concentrated to get 6.8 g product, with a yield of 85%.

Synthesis of Intermediate 1-2

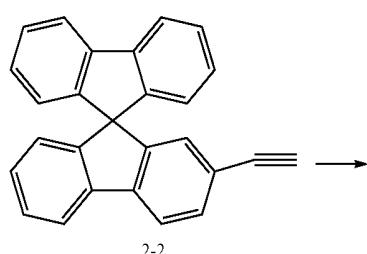
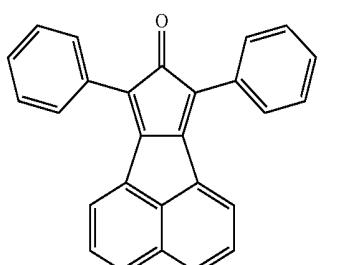
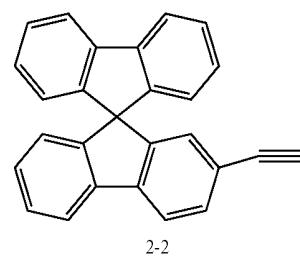
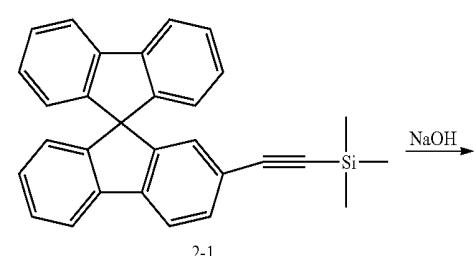
[0031] 45 ml methanol, 50 ml methylene chloride, 5 g potassium hydroxide, 6.8 g intermediate 1-1 were added to a flask under nitrogen and stirred for 1 hour. The mixture is filtered to remove the inorganic salts, and the organic solvent was removed under reduced pressure. The precipitate was recrystallized from methanol to get 4.2 g product, with a yield of 82%.

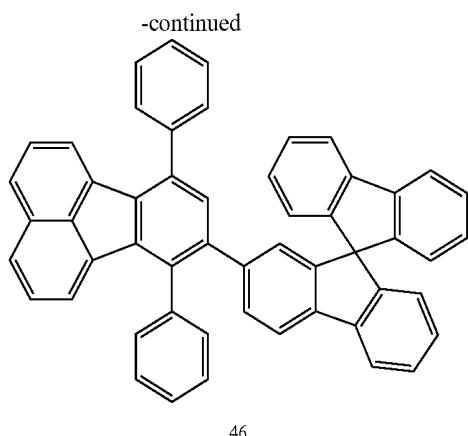
Synthesis of Intermediate 1-3

[0032] 84 g acenaphthene-quinone, 72.8 g 1,3-diphenyl acetone, 600 ml ethanol, 56 g potassium hydroxide were added into a four-necked flask under nitrogen and stirred with refluxing for 2 hours. The mixture was cooled down to the room temperature and filtered. The precipitate was rinsed with ethanol twice to get 130 g black solid, with a yield of 91%.

Synthesis of Compound 43

[0033] 4.2 g Intermediate 1-2, 6.3 g intermediate 1-3 and 60 ml diphenyl ether were added into a four-neck flask under nitrogen and stirred for 10 min, then heated to reflux for 12 hours. The mixture was cooled down and filtered. The precipitate was then stirred in refluxing ethyl acetate, cooled down and filtered to get 6.5 g light yellow solid, with a yield of 66%. <sup>1</sup>H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>) δ: 7.56-7.74 (m, 9H), 7.21-7.44 (m, 14H), 7.76-7.70 (m, 1H), 1.27 (s, 6H). MALDI-TOF-MS m/z found 546.5; C<sub>43</sub>H<sub>30</sub> ([M<sup>+</sup>]) requires 546.2.





## Synthesis of Intermediate 2-1

[0035] Triethylamine was added into a mixture of 8.0 g intermediate 2-0, 0.15 g cuprous iodide and 0.3 g tetrakis (triphenylphosphine)palladium under nitrogen. The starting materials were stirred until fully dissolved followed by adding 10 ml trimethylsilyl acetylene. The mixture was refluxed for 12 hours overnight, and the solvent was removed under reduced pressure. The residue was extracted three times using diethyl ether/water. The organic phase was washed three times with saturated brine, dried, filtered, and concentrated to get 6.8 g product, with a yield of 83%.

## Synthesis of Intermediate 2-2

[0036] 45 ml methanol, 50 ml methylene chloride, 5 g potassium hydroxide, 7 g intermediate 2-1 were added into a flask under nitrogen and stirred for 1 hour. Then the mixture was filtered to remove the inorganic salts, and the organic solvent was removed under reduced pressure. The precipitate was recrystallized from methanol to get 4.0 g product, with a yield of 70%.

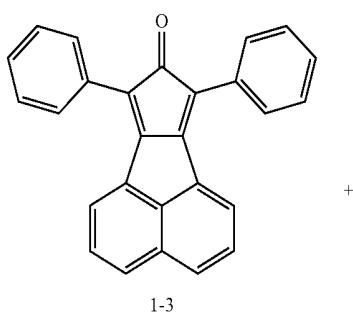
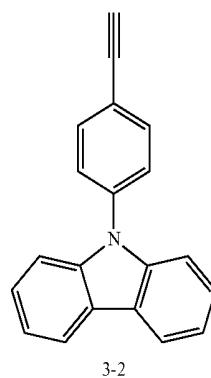
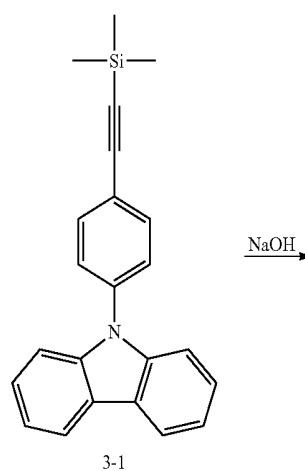
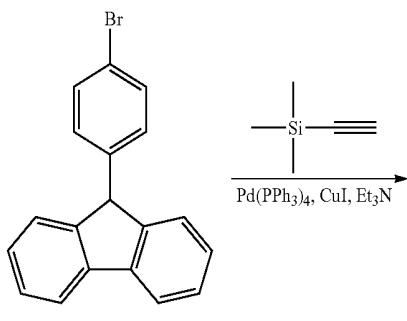
## Synthesis of Compound 46

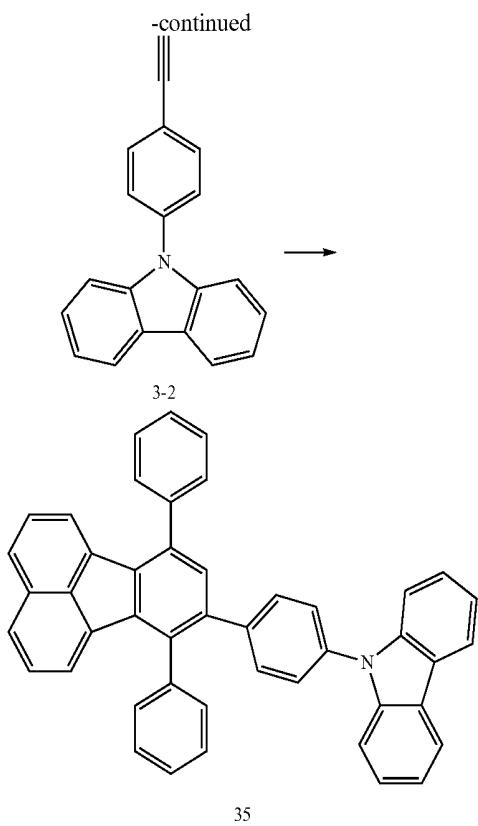
[0037] 4.0 g Intermediate 2-2, 4.2 g intermediate 1-3 and 60 ml diphenyl ether were added into a four-neck flask under nitrogen. The mixture was stirred for 10 min before refluxing for 12 hours. The mixture was cooled down and filtered. The precipitate was washed in refluxing ethyl acetate, cooled down and filtered to get 7.0 g light yellow solid, with a yield of 89%. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ: 7.61-7.79 (m, 8H), 7.47-7.52 (m, 3H), 7.32-7.36 (m, 5H), 7.20-7.24 (m, 3H), 7.02-7.14 (m, 8H), 6.56-6.65 (m, 4H), 6.49 (s, 1H), as shown in FIG. 2. MALDI-TOF-MS m/z found 668.5, C<sub>53</sub>H<sub>32</sub>[M<sup>+</sup>] requires 668.3.

## Embodiment 3

## Synthesis of Compound 35

[0038]





#### Synthesis of Intermediate 3-1

**[0039]** Triethylamine was added into a mixture of 7.7 g intermediate 3-0, 0.15 g cuprous iodide, and 0.3 g tetrakis (triphenylphosphine)palladium under nitrogen. The starting materials were stirred until fully dissolved, and then 10 ml trimethylsilyl acetylene was added, followed by refluxing the mixture for 12 hours overnight. The solvent was removed under reduced pressure, and the resulting residue was extracted three times using diethyl ether/water. The organic phase was washed three times with saturated brine, dried, filtered, and concentrated to get 6.8 g product, with a yield of 67%.

#### Synthesis of Intermediate 3-2

**[0040]** 45 ml methanol, 50 ml methylene chloride, 5 g potassium hydroxide, and 5.5 g intermediate 3-1 were added into a flask under nitrogen followed by stirring for 1 hour. The mixture was filtered to remove the inorganic salts, and organic solvent was removed under reduced pressure. The precipitate was recrystallized from methanol to get 4.11 g product, with a yield of 95%.

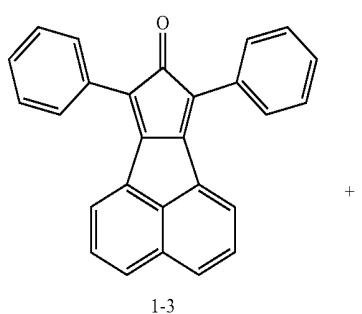
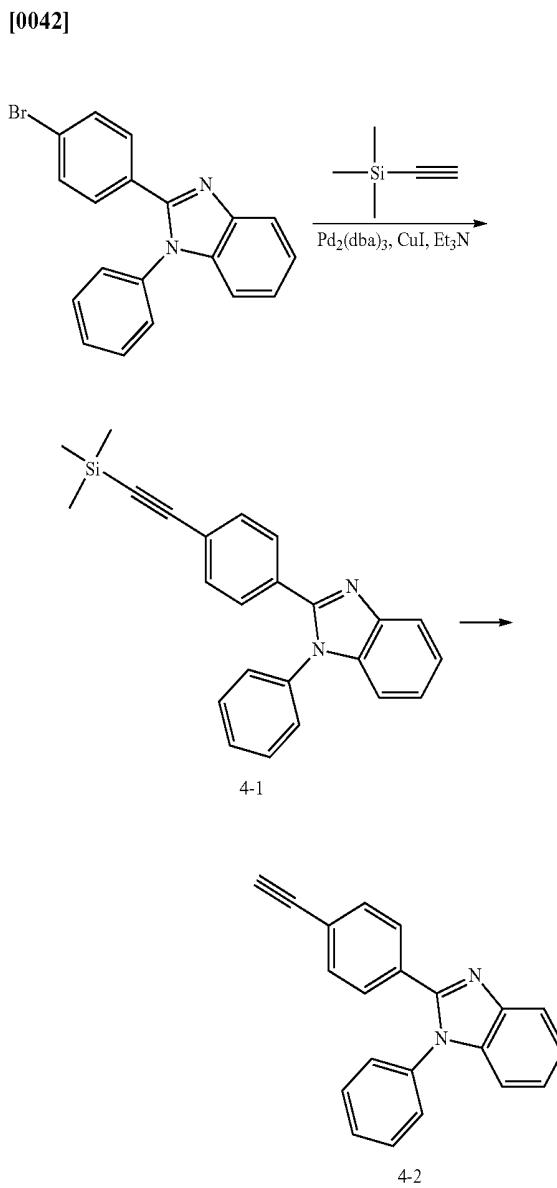
#### Synthesis of Compound 35

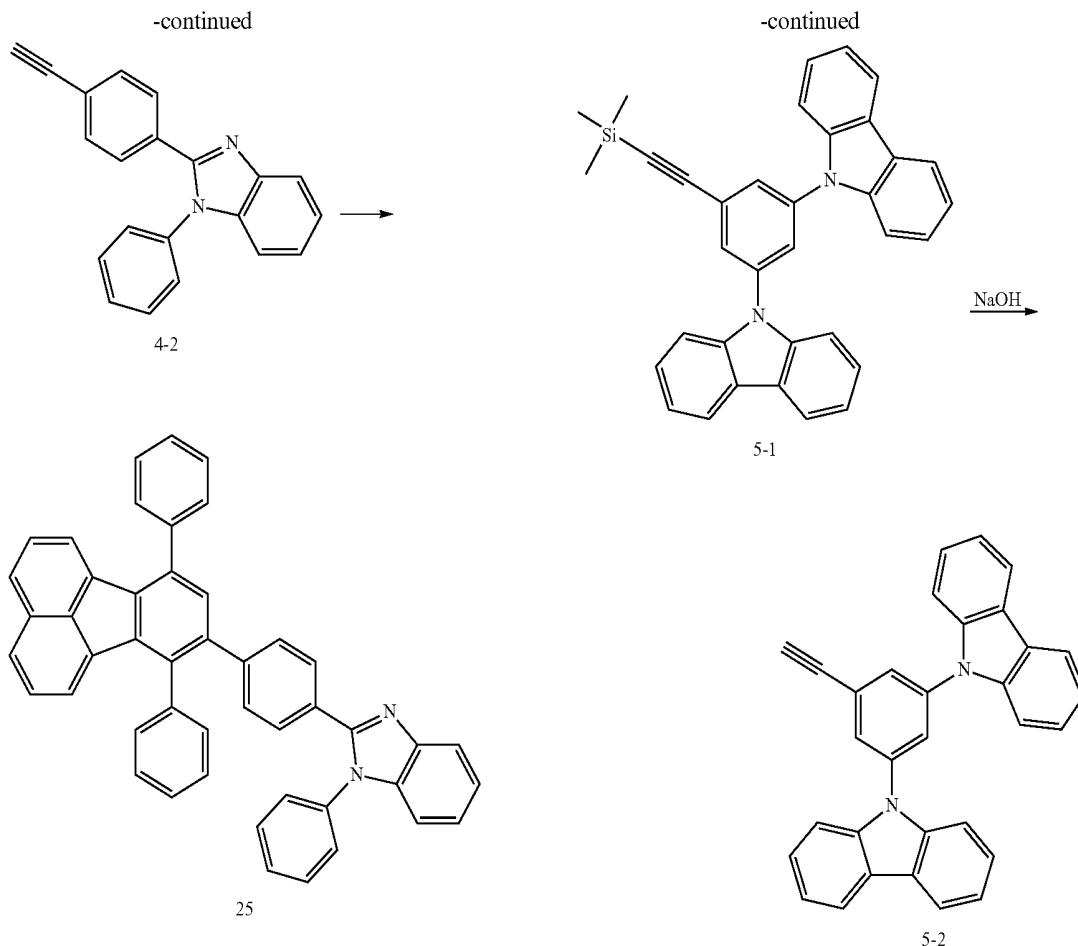
**[0041]** 4.0 g Intermediate 3-2, 5.5 g intermediate 1-3 and 60 ml diphenyl ether were added into a four-neck flask under nitrogen, and stirred for 10 min before heated to reflux for 12 hours. The mixture was cooled down and filtered. The precipitate was washed in refluxing ethyl acetate, then cooled down and filtered to get 4.5 g light yellow solid, with a yield of 49%.  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_2\text{Cl}_2$ )  $\delta$ : 8.12-8.13 (d,  $J=7.6$  Hz, 2H), 7.74-7.77 (m, 4H), 7.52-7.60 (m, 3H), 7.25-7.45 (m,

19H), 6.69-6.81 (d,  $J=7.2$  Hz, 1H). MALDI-TOF-MS  $m/z$  found 595.4;  $\text{C}_{46}\text{H}_{29}\text{N}$  ( $[\text{M}^+]$ ) requires 595.2.

#### Embodiment 4

##### Synthesis of Compound 26



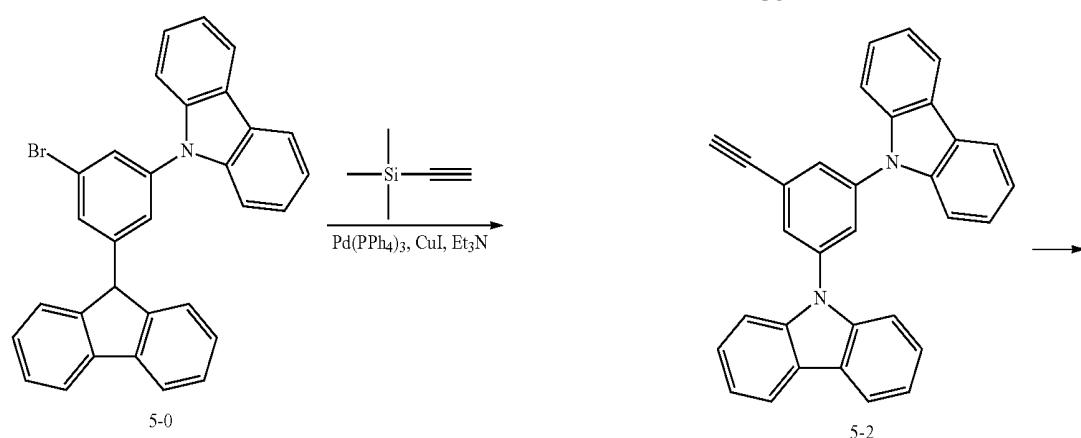


[0043] The synthesis process is the same as that of embodiment 1

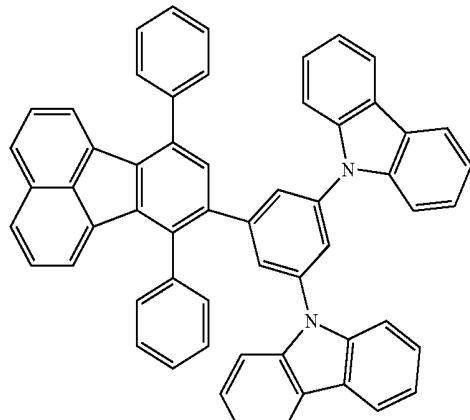
Embodiment 5

Synthesis of Compound 36

[0044]



-continued



36

## Synthesis of Intermediate 5-1

[0045] 100 ml triethylamine was added into a mixture of 9.1 g compound 5-0, 0.15 g cuprous iodide, and 0.24 g tetrakis(triphenylphosphine)palladium under nitrogen. The starting materials were stirred until fully dissolved, and then 10 ml trimethylsilyl acetylene was added, followed by refluxing for 12 hours overnight. The solvent was removed under reduced pressure, and the resulting residue was extracted three times using diethyl ether/water. The organic phase was washed three times with saturated brine, dried, filtered, and concentrated to get 6.8 g product, with a yield of 74%.

## Synthesis of Intermediate 5-2

[0046] 45 ml methanol, 50 ml methylene chloride, 5 g potassium hydroxide, and 7 g intermediate 2-1 were added into a flask under nitrogen followed by stirring for 1 hour. The mixture was filtered to remove the inorganic salts, and the organic solvent was removed under reduced pressure. The precipitate was recrystallized from methanol to get 5.0 g product, with a yield of 83%.

## Synthesis of Compound 36

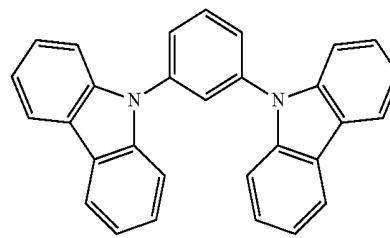
[0047] 4.8 g intermediate 6-2, 3.9 g intermediate 1-3 and 60 ml diphenyl ether were added into a four-neck flask under nitrogen. The mixture was stirred for 10 min before heated to reflux for 12 hours. The mixture was cooled down and filtered. The precipitate was washed in refluxing ethyl acetate, cooled down and filtered to get 5.5 g light yellow solid, with a yield of 66%. <sup>1</sup>H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>) δ: 8.14-8.16 (d, J=8.4 Hz, 4H), 7.72-7.80 (m, 4H), 7.54-7.69 (m, 11H), 7.27-7.50 (m, 16H), 6.74-6.76 (d, J=6.8 Hz, 1H), as shown in FIG. 3. MALDI-TOF-MS m/z found 760.5; C<sub>59</sub>H<sub>36</sub>N<sub>2</sub> ([M<sup>+</sup>]) requires 760.3.

## Embodiment 6

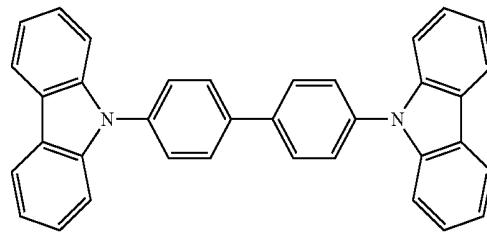
[0048] With the protection of nitrogen, the glass transition temperature of the compound 26 was measured using DSC method at a heating and cooling rate of 20° C./min. The measured glass transition temperature T<sub>g</sub> of Compound 26 is 140° C. The glass transition temperatures of compounds 35, 36, 43, 46 are measured using the same procedures. The results are listed in the following table 1.

## Comparison Example 1

[0049] The glass transition temperatures of compounds mCP and BCP were measured using the same test procedures as stated in Embodiment 6. The results are also listed in the following table 1.



mCP



BCP

	Compound	Glass transition temperature (T <sub>g</sub> )° C.
Embodiment	Compound 26	140
	Compound 35	132
	Compound 36	169
	Compound 43	90
	Compound 46	167
Comparison Example	mCP	55
	BCP	62

[0050] As shown from table 1, the compounds in the present invention have higher glass transition temperatures than some conventional host materials in the market. This invention has improved the thermal stability of the electroluminescent materials.

## Embodiment 7

[0051] OLED was fabricated adopting the organic electronic material in the present invention. The structure of device is shown in FIG. 1.

[0052] Firstly, the transparent and conductive ITO substrate (with the ITO layer 20 above the glass 10) was washed with detergent solution, deionized water, ethanol, acetone, deionized water in sequence, then treated with oxygen plasma for 30 seconds, and then treated with CF<sub>x</sub> plasma.

[0053] Then on ITO was vacuum evaporated 5 nm MoO<sub>3</sub> as the hole injection layer 30.

[0054] Then 50 nm thick TAPC was vacuum evaporated as the hole transport layer 40.

[0055] Then on top of the hole transport layer was evaporated 10 nm light emitting layer, which comprises compound 35 as a host material and 10% FIrpic as a blue phosphorescent material.

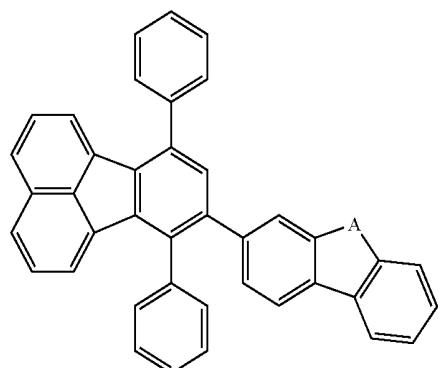
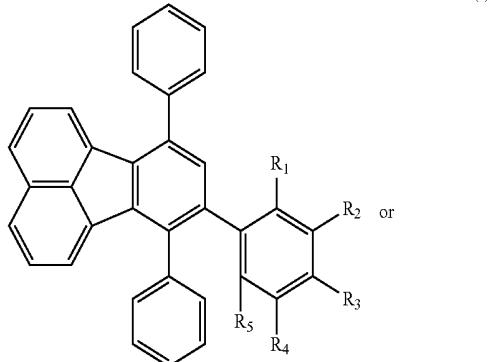
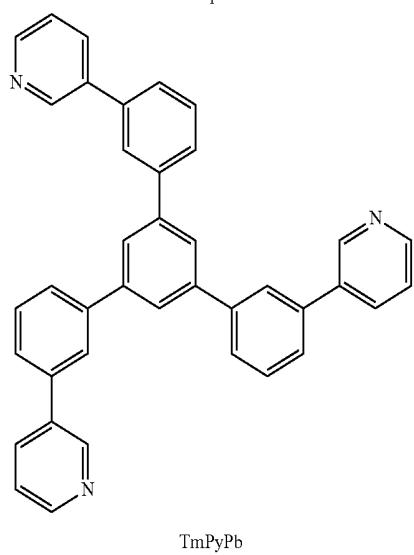
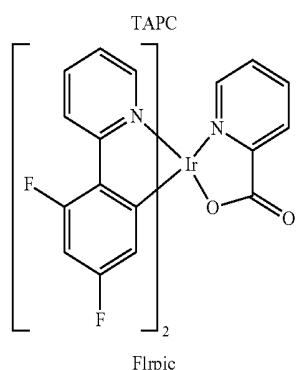
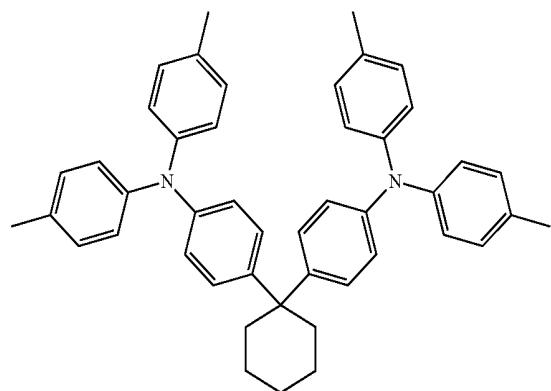
[0056] Then, 50 nm thick TmPyPb was vacuum-evaporated as an electron transport layer 60 on top of the light emitting layer.

[0057] Finally, 1.2 nm LiF, 150 nm Al were evaporated as the electron injection layer 70 and the cathode, respectively.

**[0058]** The OLED produced as mentioned above can achieve a current density of  $0.28 \text{ mA/cm}^2$  under a driving voltage of 4V with blue emission.

voltage of 4V with blue emission. Therefore, compared with conventional materials, the materials in the present invention can give higher current density at the same driving voltage, improving the light-emitting performance.

1. An organic electronic material having the structure of formula (I) or (II) as described below:



wherein, any four of  $R_1$ - $R_5$  are hydrogen, and the other is C1-C6 alkyl, C7-C24 aryl, substituted or unsubstituted heteroaromatic ring(s) having one or more heteroatoms (N, O, S), triaryl silyl, diarylamino, diaryl phosphine oxide, aromatic carbonyl, arylthio; or

$R_1, R_3, R_5$  are hydrogen,  $R_2, R_4$  independently represent C1-C6 alkyl, C6-C24 aryl, heteroaromatic ring(s) with one or more heteroatoms (N, O, S), triaryl silyl, diaryl amino, diaryl phosphine oxide, aromatic carbonyl, arylthio,

A represents C(R<sub>6</sub>)<sub>2</sub>, N(R<sub>7</sub>), S, O, P(R<sub>8</sub>), S(O)<sub>2</sub> or B(R<sub>9</sub>), R<sub>6</sub>-R<sub>9</sub> independently represent hydrogen, deuterium, alkyl, phenyl, alkylphenyl, heteroaromatic ring with one or more heteroatoms (N, O, S), a cyclized structure formed between two R<sub>6</sub> and C.

2. The organic electronic material according to claim 1, wherein any four of  $R_1$ - $R_5$  are hydrogen and the other is five-membered or six-membered heteroaromatic ring substituted with fluorenyl or alkyl or phenyl or naphthyl, diphenyl amino, phenyl-naphthylamine, triphenylsilyl, diphenyl phosphine oxide, phenyl carbonyl or phenyl sulfonyl.

3. The organic electronic material according to claim 2, wherein the said five-membered or six-membered heteroaromatic ring is carbazolyl, pyrimidinyl, pyridyl, thiazolyl, triazole group or triazine group.

### Comparison Example 2

[0059] The device structure was the same as that in Embodiment 7, but the compound 35 was replaced by CBP.

[0060] The OLED produced in the comparison example can achieve a current density of  $0.17 \text{ mA/cm}^2$  under a driving

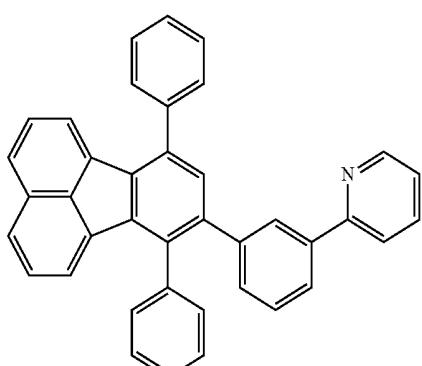
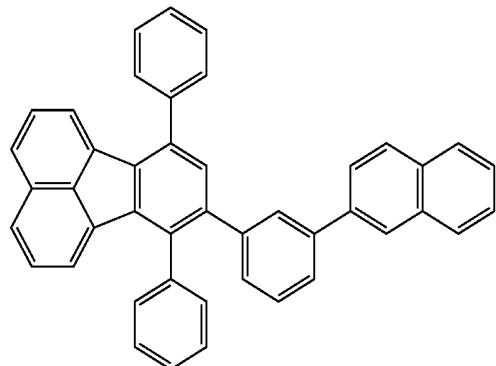
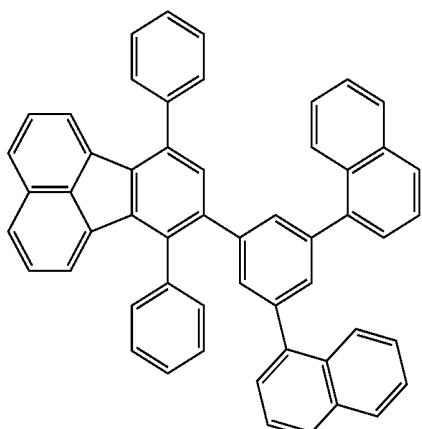
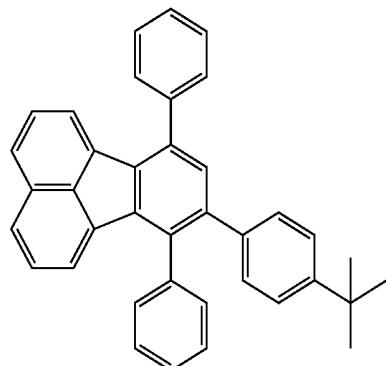
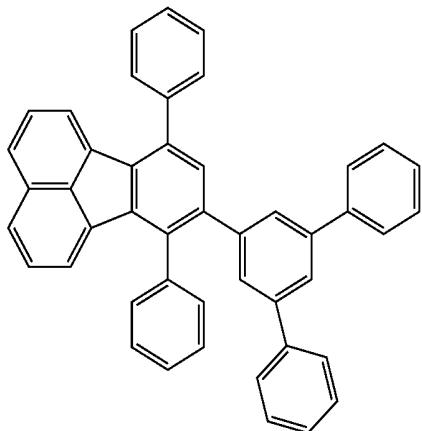
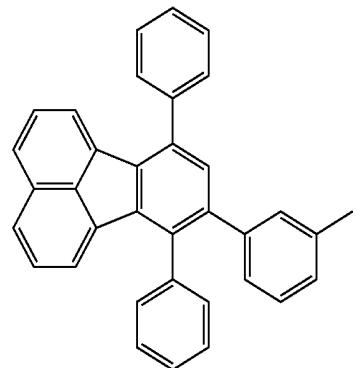
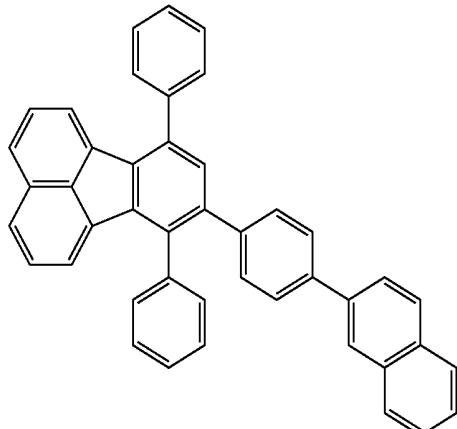
4. The organic electronic material according to claim 2, wherein the said fluorenyl is 9,9-dimethyl-fluorenyl, 9,9-diphenyl fluorenyl, 9,9-di-p-tolyl-9H-fluorenyl or spiro fluorenyl.

5. The organic electronic material according to claim 2, wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>4</sub>, R<sub>5</sub> are hydrogen.

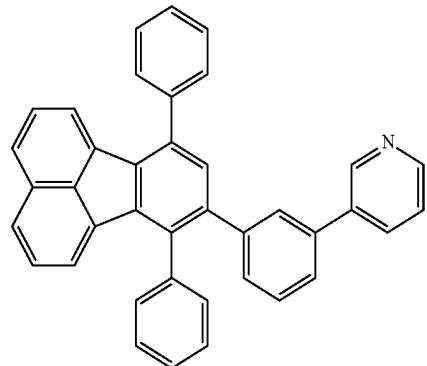
6. An organic light-emitting material according to claim 1, wherein A is C(R<sub>6</sub>)<sub>2</sub>, N(R<sub>7</sub>), S, O, S(O)<sub>2</sub>; R<sub>6</sub>, R<sub>7</sub> independently represent hydrogen, methyl, phenyl, or methyl phenyl, or five-membered cyclic structure formed between two R<sub>6</sub> and C.

7. The organic electronic material according to claim 1, wherein its structure is as below:

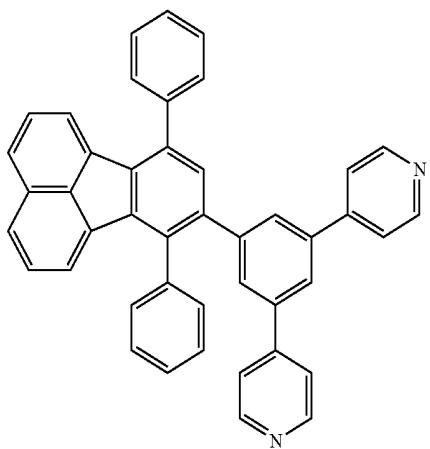
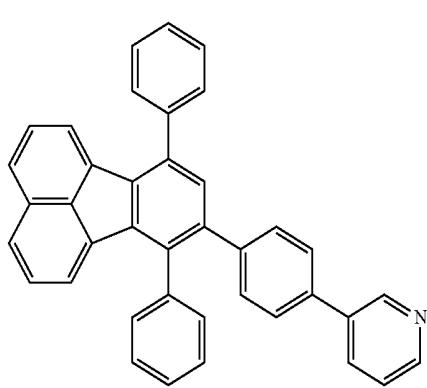
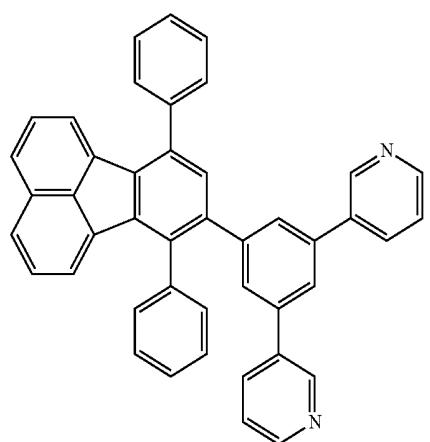
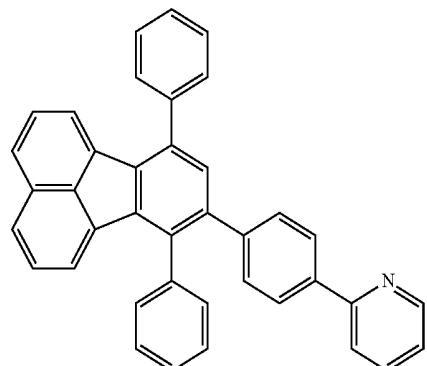
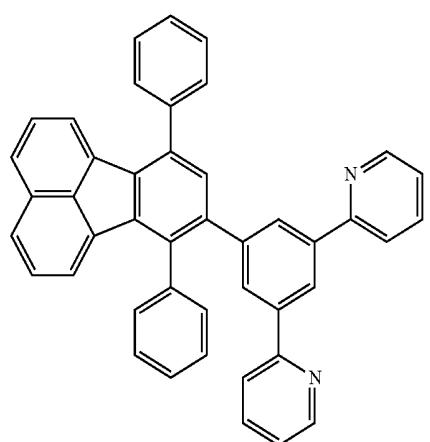
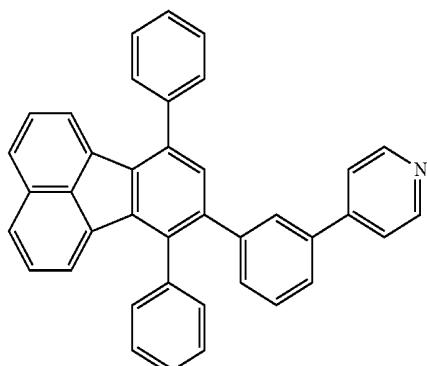
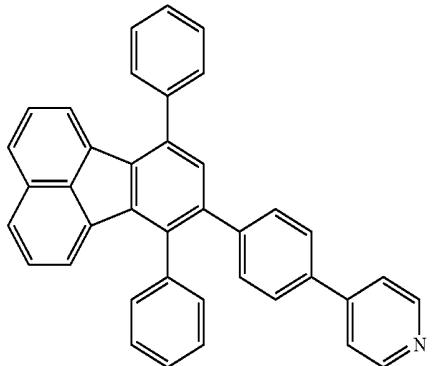
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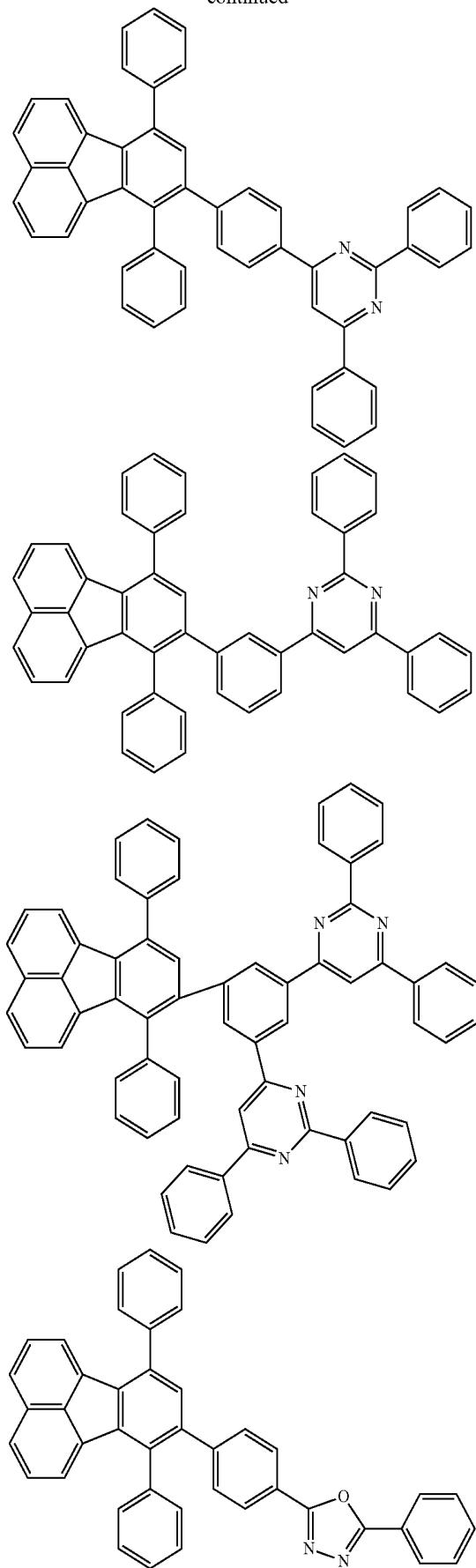
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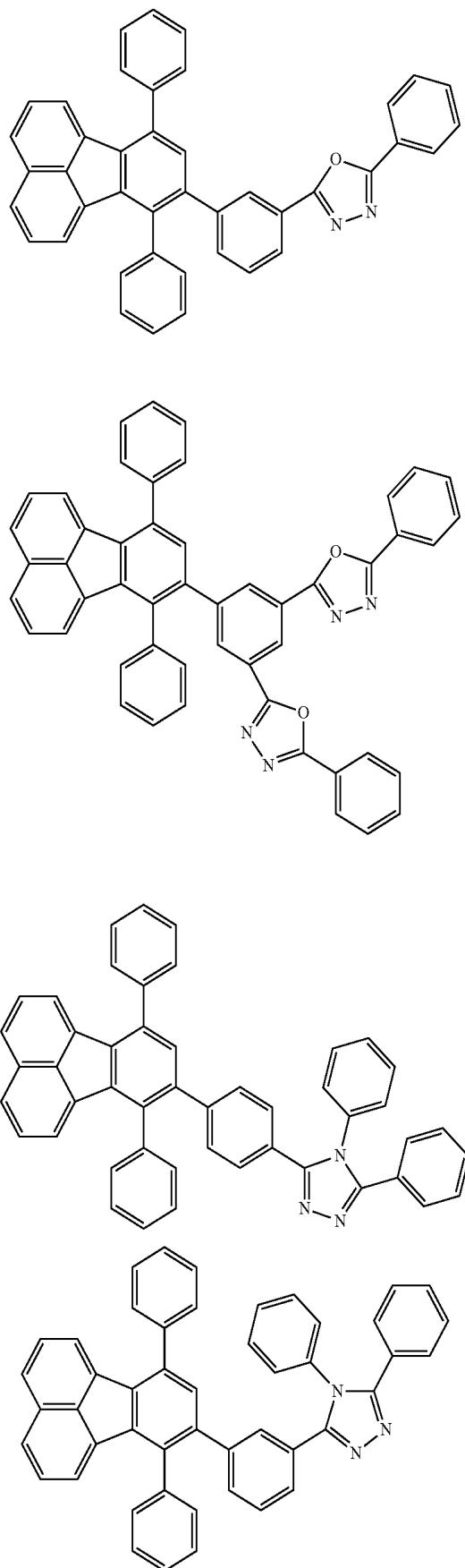
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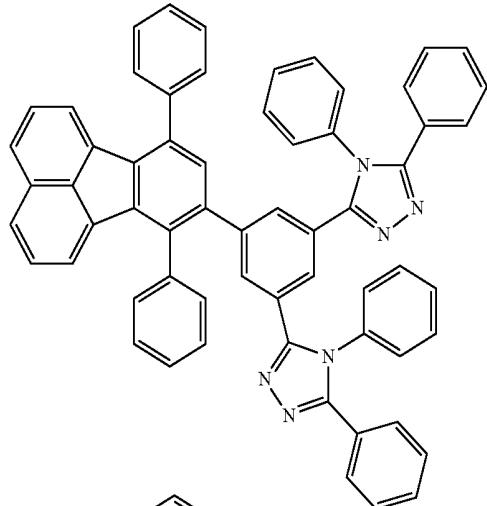
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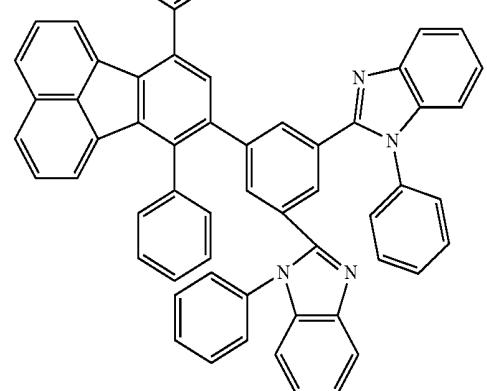
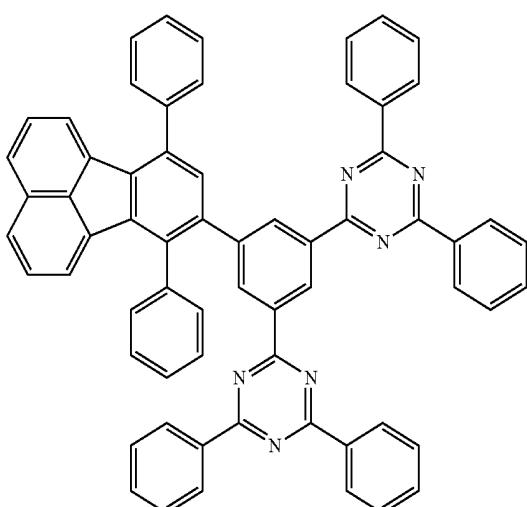
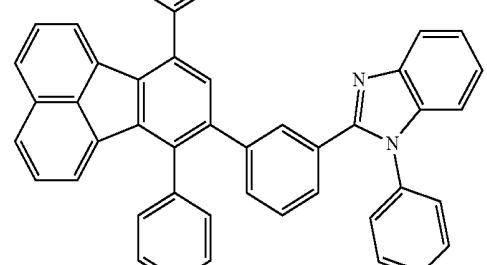
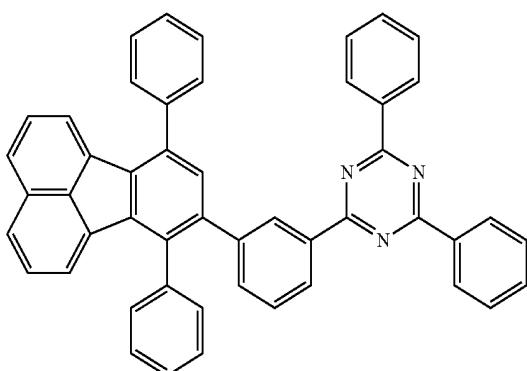
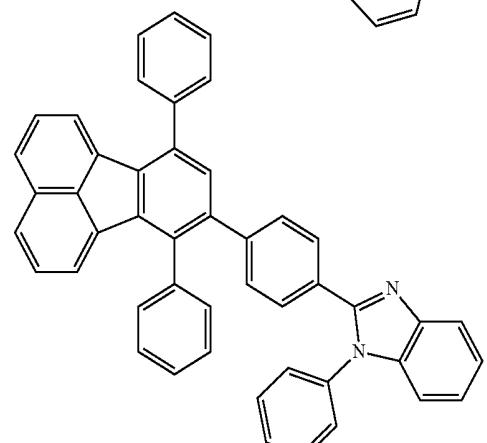
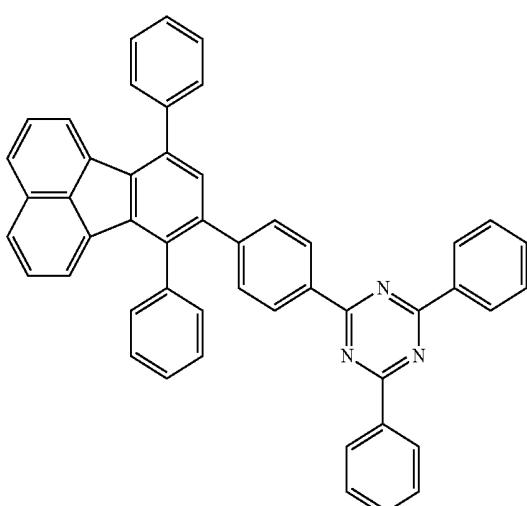
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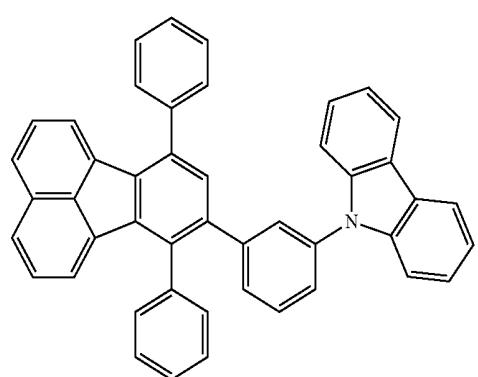
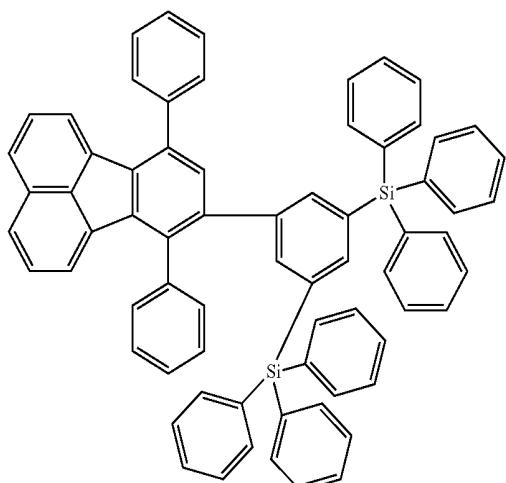
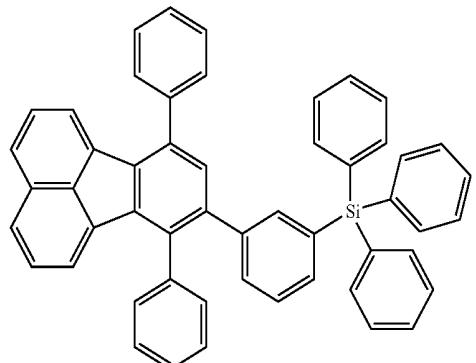
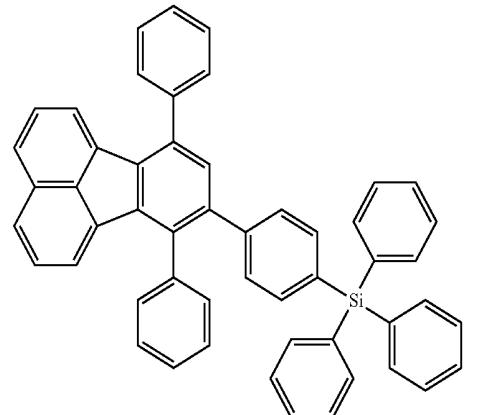
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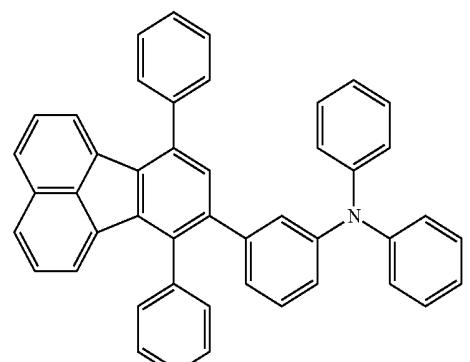
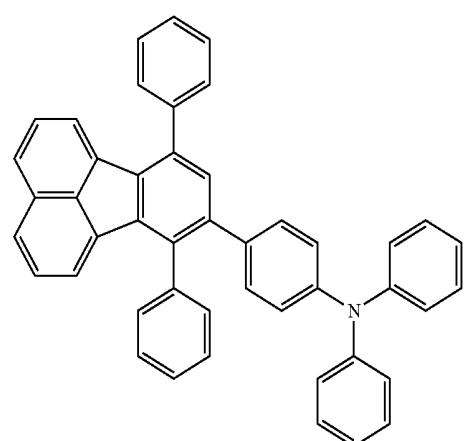
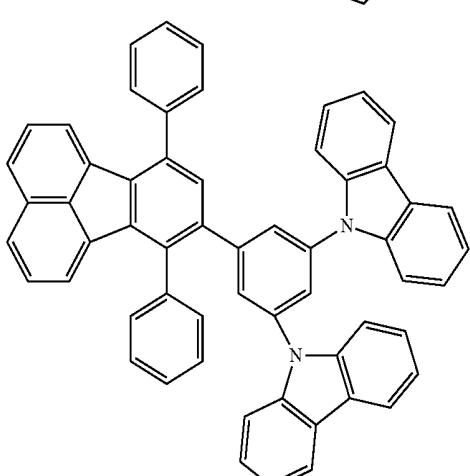
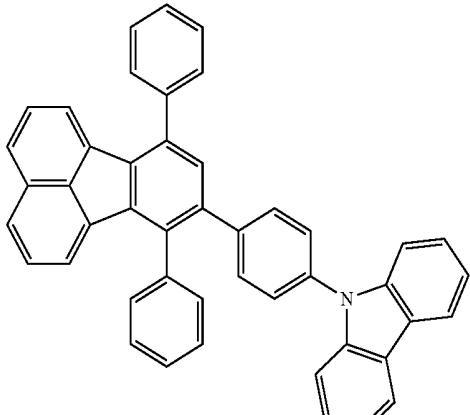
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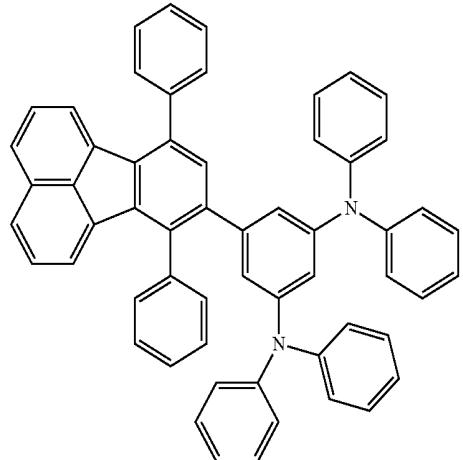
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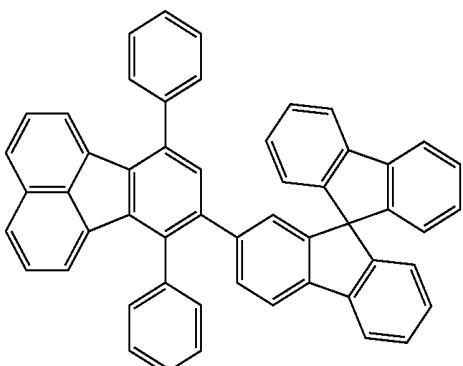
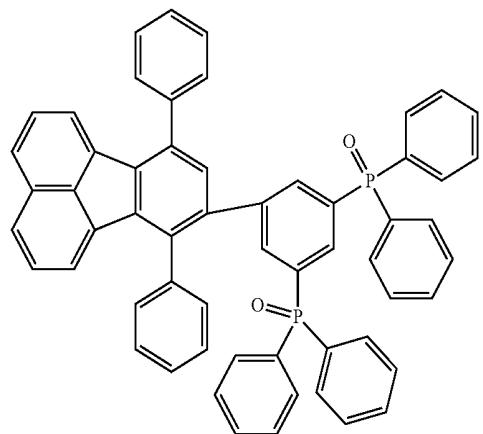
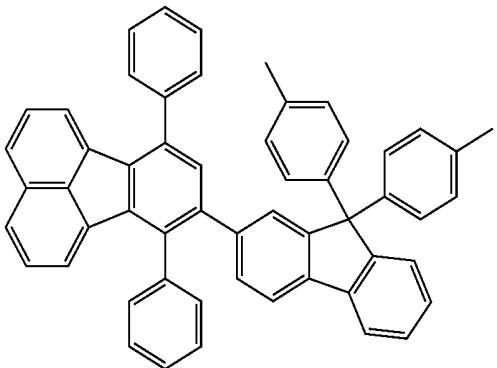
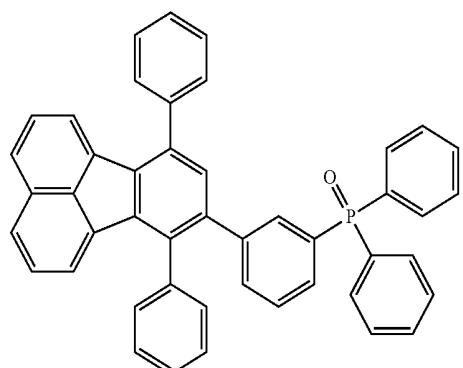
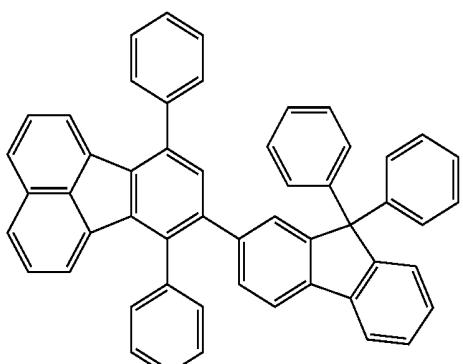
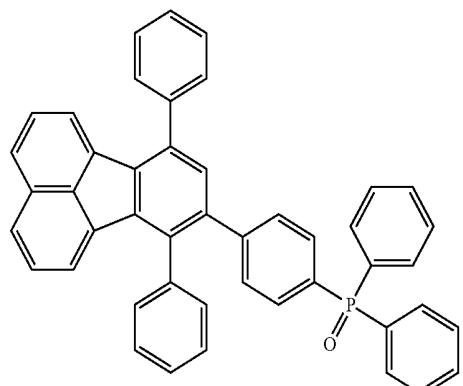
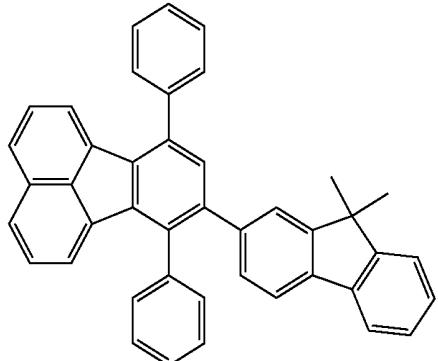
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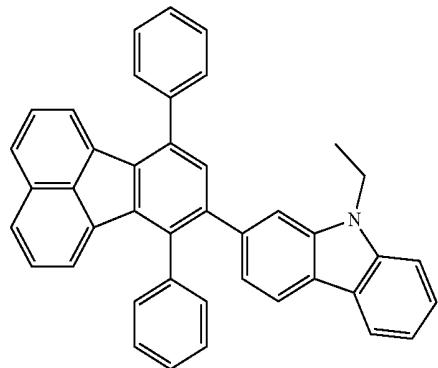
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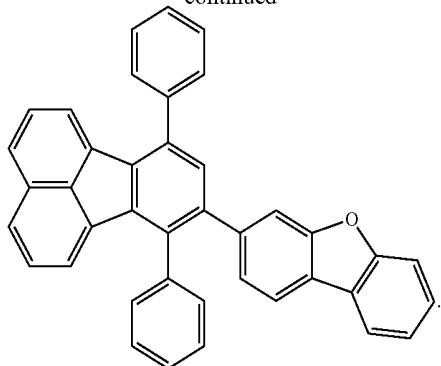
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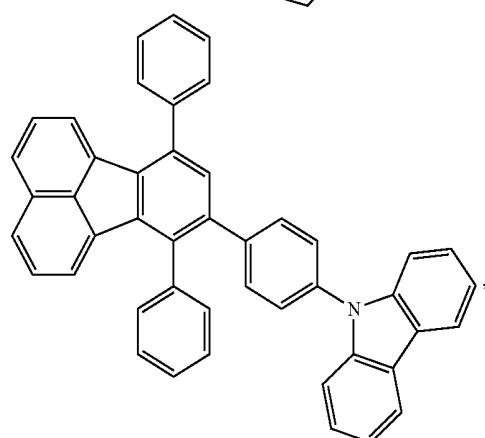
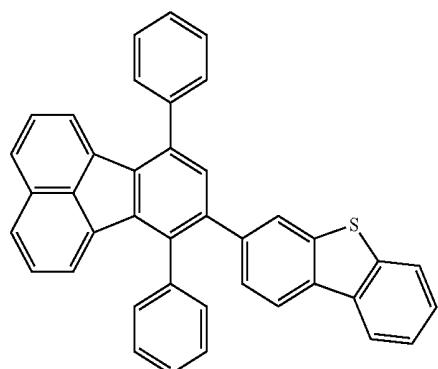
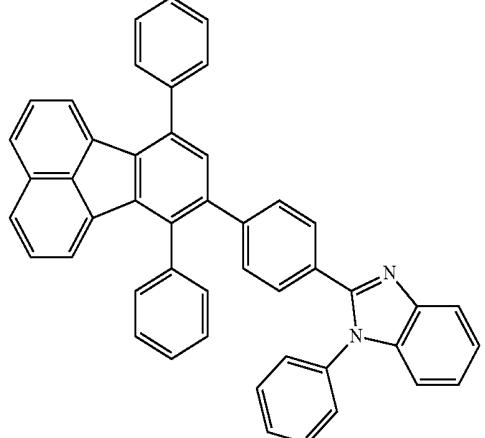
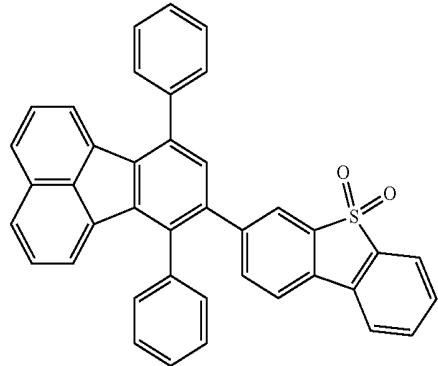
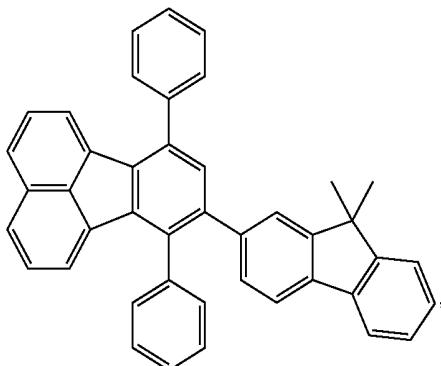
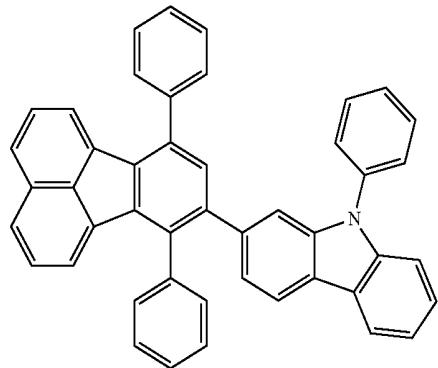
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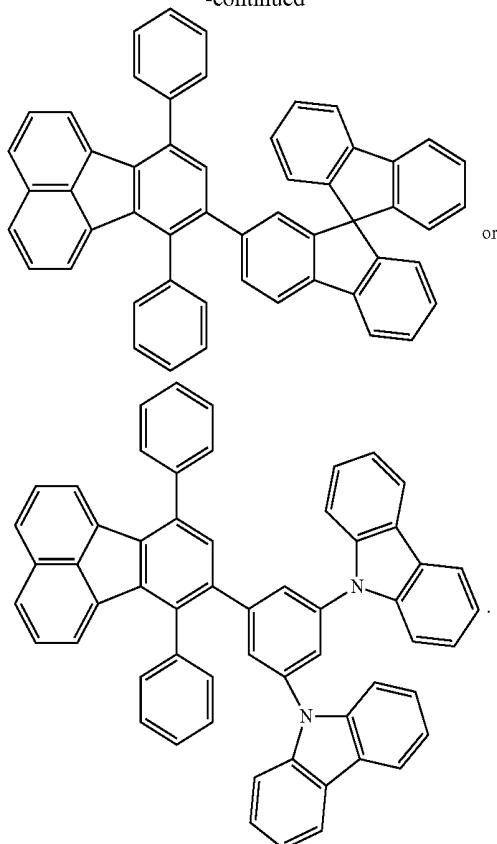
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8. The organic electronic material according to claim 1, wherein its structural formula is as below:



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**9.** An OLED containing the organic electronic material as stated in claim 1.

**10.** The OLED according to claim 9, comprising a hole injection layer, a hole transport layer, a light emitting layer, an electron transport layer or/and an electron injection layer, wherein, the organic electronic material as stated in claim 1 is used in any one layer or multiple layers of hole injection layer, hole transport layer, light emitting layer, electron transport layer or/and electron injection layer.

**11.** The OLED according to claim 10, wherein the organic electronic material is used in the electron transport layer.

**12.** The OLED according to claim 10, wherein the said light-emitting layer comprises a host material and a guest material, and the organic electronic material is the host material for red phosphorescent dopant in the light-emitting layer.

**13.** The OLED according to claim 12, wherein the guest material is an organic iridium compound or an organic platinum compound.

**14.** The OLED according to claim 12, wherein the organic electronic material is used in the electron transport layer.

**15.** The OLED according to claim 13, wherein the organic electronic material is used in the electron transport layer.

**16.** The organic electronic material according to claim 3, wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>4</sub>, R<sub>5</sub> are hydrogen.

**17.** The organic electronic material according to claim 4, wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>4</sub>, R<sub>5</sub> are hydrogen.

\* \* \* \* \*

专利名称(译)	有机电子材料和有机电致发光器件		
公开(公告)号	<a href="#">US20150108449A1</a>	公开(公告)日	2015-04-23
申请号	US14/404609	申请日	2013-06-05
[标]申请(专利权)人(译)	广东阿格蕾雅光电材料 北京阿格蕾雅科技发展有限公司		
申请(专利权)人(译)	广东树兰光电子材料有限公司 北京树兰科技发展有限公司		
当前申请(专利权)人(译)	广东树兰光电子材料有限公司 北京树兰科技发展有限公司		
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IPC分类号	H01L51/00 C07D235/08 C07C13/66 C07D209/86		
CPC分类号	H01L51/0054 C07D209/86 C07D235/08 C07C13/66 H01L51/0058 C07C2103/40 H01L51/5072 H01L51/5012 H01L51/5056 H01L51/5092 H01L51/0072 C07C13/72 C07C211/54 C07C2603/18 C07C2603/40 C07C2603/94 C07D213/06 C07D213/16 C07D235/18 C07D235/20 C07D239/26 C07D249/08 C07D251/24 C07D271/107 C07D307/91 C07D333/76 C07F7/0805 C07F9/5325 C07F9/5329 C07F15/0033 C09B57/00 C09B57/008 C09B69/008 C09K11/06 C09K2211/1007 C09K2211/1011 C09K2211/1014 C09K2211/1029 C09K2211/1044 C09K2211/1048 C09K2211/1051 C09K2211/1059 H01L51/0067 H01L51/0094 H01L51/5016		
优先权	201210185154.3 2012-06-06 CN		
其他公开文献	US10026901		
外部链接	<a href="#">Espacenet</a> <a href="#">USPTO</a>		

**摘要(译)**

公开了具有如下所述的式(I)或(II)的结构的有机发光材料和有机发光装置(OLED)。OLED采用含荧蒽基的化合物作为电子传输材料,具有良好的电子传输和注入能力。由于其优异的热稳定性和成膜性,该材料还提高了器件的发光效率和寿命。同时,含有荧蒽基团的材料的高三重态能量和优异的电子传输能力使其适合用作磷光装置的主体,增加发光层中的电子数量和装置的效率。。

